Screening and selection of a lead phytochemical inhibitor against Nsp3 protein, from *Lawsonia inermis* L. through *in silico* approaches

Debapriya Das[1], Sujata Dhar Dutta[2], Bidisha Mallick[1], Dipu Samanta[3] & Debleena Roy[1*]

[1]Post Graduate Department of Botany, Lady Brabourne College, P-1/2 Suhrawardy Avenue, Beniapukur, Kolkata 700 017, West Bengal, India
[2]Department of Zoology, Lady Brabourne College, P-1/2 Suhrawardy Avenue, Beniapukur, Kolkata 700 017, West Bengal, India
[3]Department of Botany, Dr. Kanailal Bhattacharyya College, Howrah 711 104, West Bengal, India

*Correspondence email - debleena.roy@ladybrabourne.com

Abstract

Lawsonia inermis L., a plant of immense significance as the natural source of dye "*Mehendi*" is rich in secondary metabolites that impart important medicinal properties. These phytochemicals offer plant-based alternatives to synthetic drugs, which are often associated with harmful side effects, reinforcing the need to explore such natural compounds. The main objective of this study is to find plant-based lead from among the phytochemicals present in *Lawsonia inermis* L., against Nsp3 protein. Methanolic extracts from the plant was used and a group of phytochemicals were verified and identified by ultra-high-performance liquid chromatography coupled with quadrupole time-of-flight mass spectrometry, which were then subjected to phytochemical screening. The shortlisted best ligands (caffeic acid and lawsone) were subjected to molecular dynamic simulations for 100 ns, where RMSD, Rg and RMSF were analysed along with the binding affinities to the protein. Lawsone and caffeic acid emerged as good candidates. Molecular dynamic simulations indicated that lawsone, a naphthoquinone, formed stable complexes with Nsp3, exhibiting favourable RMSD, RMSF and Rg values, as well as a strong binding affinity of -20.72 kcal/mol. Our analysis shows the prospective possibility of lawsone, as a potential antiviral compound. This also highlights the interactions of a naphthoquinone, against Nsp3 protein, increasing the medicinal importance of the plant. Previous studies usually reveal phenols and flavonoids to be effective antiviral agents. This study, for the very first time, reveals the potential of lawsone (naphthoquinone and the main pigment of the plant), against Nsp3 protein.

Keywords: docking; *Lawsonia inermis*; molecular dynamics; Nsp3; phytochemicals; screening

Introduction

India with its rich diversity of medicinal plants stands tall among the countries which have a long history of plant-based medicines (1-4). Ayurveda which originated around 5000 years ago is gradually witnessing an era of commercialization, with a rise in demand for alternatives to diseases, with less harmful side-effects (5,6). The market of medicinal plants in India, is expected to value at around 8 billion US dollars by 2022. A knowledge and practise which originated in India, is slowly spreading globally, providing potential cure and treatment to many (7). Dravyaguna, a part of Ayurveda dealing with the properties and uses of medicinal plants, has mention of Madyantika, or *Lawsonia inermis*, being useful in the commercial and medicinal sectors (8). Some of the important medicinal properties of *Lawsonia inermis*, include, anti-carcinogenic, immune-stimulatory, antidiabetic, antiproliferative and many more (9-11). This can be attributed to the bioactive compounds present in the plant, especially which are rich in phenols and flavonoids (12,13).

Herbal medicines use a strong combination of bioactive compounds, which have less toxic side effects, providing therapeutic benefits with reduced toxicity (14,15). Although, vaccines serve to be effective, there have been reports of many side effects and they are comparatively costly (16,17). Natural bioactive compounds (phenols and flavonoids) serve as effective antiviral agents by hindering viral replication, translation, transcription, viral packaging and assembly (18,19). Quercetin, apigenin, naringenin, curcumin, chebulagic acid and punicalagin have shown effective action against viruses like HIV, DENV, Influenza and others (20,21). The antiproliferative potential of polyphenols is of utmost importance, as they have been reported to inhibit a wide range of viruses like, influenza virus, coronavirus, rhinovirus, syncytial virus, rotavirus, hepatitis virus, herpes virus, dengue virus and many more (22,23). Hence, in this study, we aim to find plant-based alternatives, from *Lawsonia inermis* L., whose mention has been found in the Ayurvedic texts (9,24). The phytochemicals identified were analysed against Nsp3, an important protein needed for the maintenance of the viral life cycle of SARS-CoV-2 (25).

Among the many important roles played by Nsp3 protein, it aids in polyprotein processing by binding to viral proteins (26). Nsp3 protein, along with Nsp4 and Nsp6 are involved in formation of Double-Membrane Vesicles (DMV) in coronavirus-infected cells, playing a key role in virus survival (27,28). Apart from this, the protein also helps in virus replication, by interacting with 3CLpro and PLpro (27,29). Methanolic extracts of *Lawsonia inermis* L., were seen as potential sanitizers, exhibiting strong antimicrobial activity, suggesting their role in bacterial and viral growth inhibition (30). The target protein Nsp3 holds immense importance in the virus survival. Hence, targeting this protein can substantially lower the pathogenicity of the virus, providing a plant-based alternative (31). Based on previous analyses, this work, for the very first time investigates the interaction of identified bioactive compounds from *Lawsonia inermis* L., against the Nsp3 protein.

Materials and Methods

Preparation of protein

Nsp3, the largest multidomain protein encoded by the virus has an average molecular weight of 200kDa (32). The structure was obtained from RCS PDB database (Fig. 1). For processing the PDB protein file, MGL tools were used, where, polar hydrogens were added, ligand and water molecules were removed before docking.

Ligand preparation

Seeds of *Lawsonia inermis* L. were obtained from the National Bureau of Plant Genetic Resources (NBPGR), New Delhi (IC 627437). Leaves of four-month-old *Lawsonia inermis* L. were dried, weighed, crushed and kept in 80 % methanol overnight for optimum extraction of phytochemicals (33). The leaf extract was subjected to Ultra-High-Performance Liquid Chromatography Quadrupole Time-of-Flight Mass Spectrometry (UHPLC–QTOF-MS) (Agilent UHPLC–QTOF-MS model 6500 series). For the process, 10 μL sample was injected, using Acquity UPLC BEH C18 column (100 mm x 2.1 mm, 1.7 μm) (34), with gradient mobile phases, A: ultrapure water (0.5 % acetic acid) and B: acetonitrile: methanol (50:50). A group of 14 phytochemicals (Fig. 2) were reported among which some were validated from the list provided in a previous work (35), along with which, some more compounds were also identified. Table 1 summarises the data from the PubChem Database.
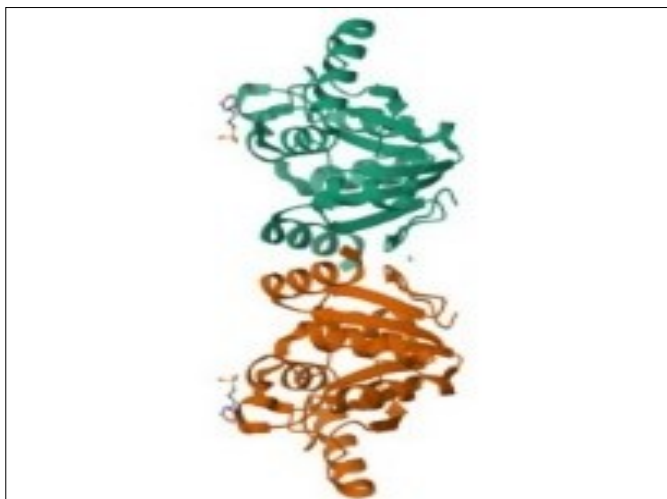

Fig. 1. Protein Nsp3 (PDB ID: 8AZC) downloaded from the RCSB PDB.

Docking analysis of ligand with the protein

Docking was done by AutoDock Vina (15), against the Nsp3 protein. Following which, the complexes were further processed for molecular dynamic simulations for 100 ns, with the CHARMM ffT92 force field (15). The log file obtained gave 9 poses for the ligand-protein complexes, along with RMSD values and binding affinity. Poses with the highest binding affinity and low RMSD values (less than 2Å) were considered (36). The interactions were recorded using the Discovery Studio (https://www.3ds.com/products/biovia/discovery-studio) and the interacting residues were noted, along with the number of hydrogen bonds.

Lipinski's rule of five

Ligands were analysed using Lipinski's rule of five, which highlights the drug "potential" of a ligand (http://www.scfbio-iitd.res.in/software/drugdesign/lipinski.jsp) (37).

In silico ADME analysis

Swiss ADME was used to study the behaviour of the ligands in the human body by, analysing characteristics such as metabolism, absorption, excretion and distribution (http://www.swissadme.ch/index.php) (38,39).

Bioactivity score

This parameter helps us to know about the interaction of the prospective drug candidates with the drug receptors present in the body, done by Molinspiration software (https://www.molinspiration.com/). A bioactivity score of 0 and more, shows good bioactivity, whereas lower scores indicate poor activity (40,41).
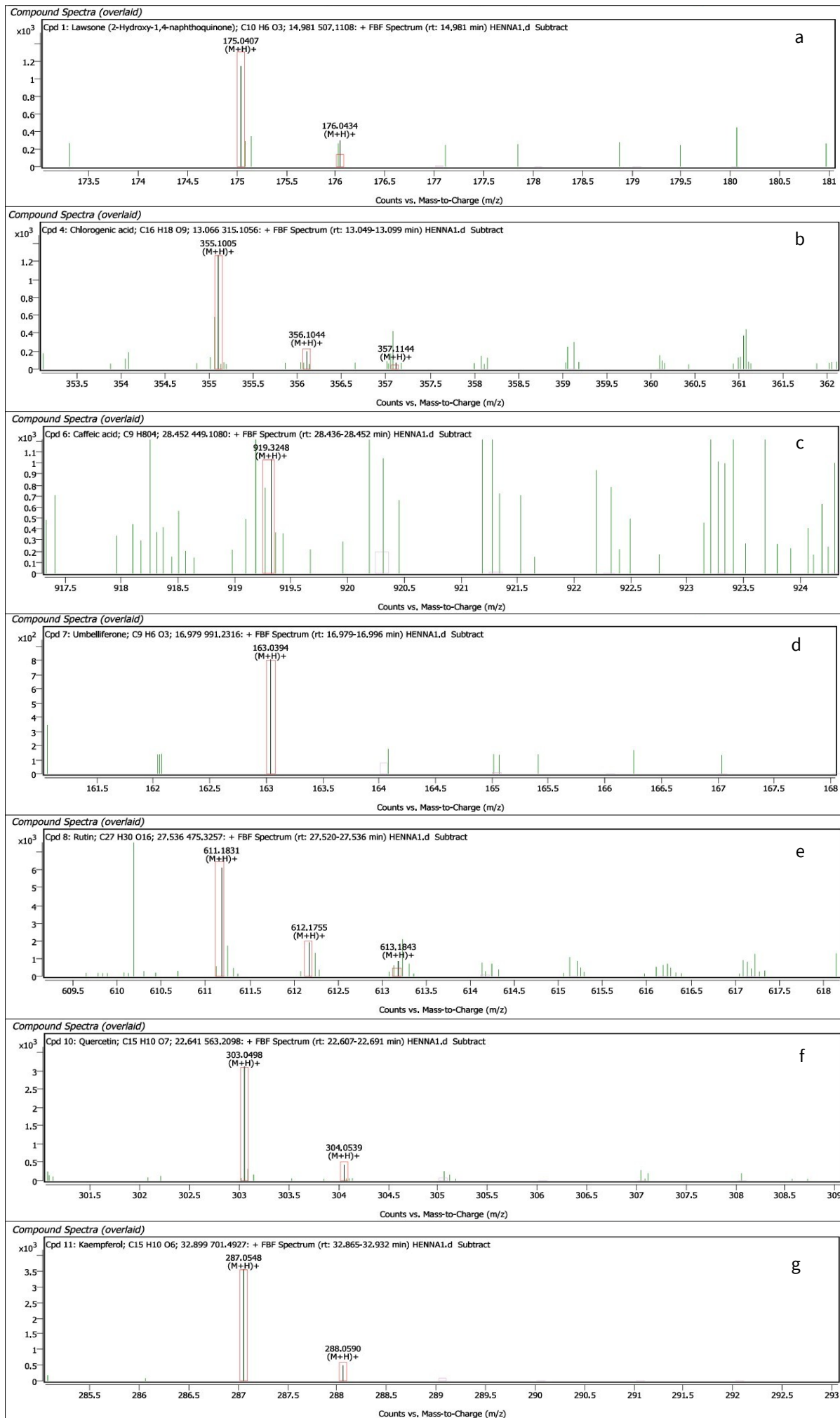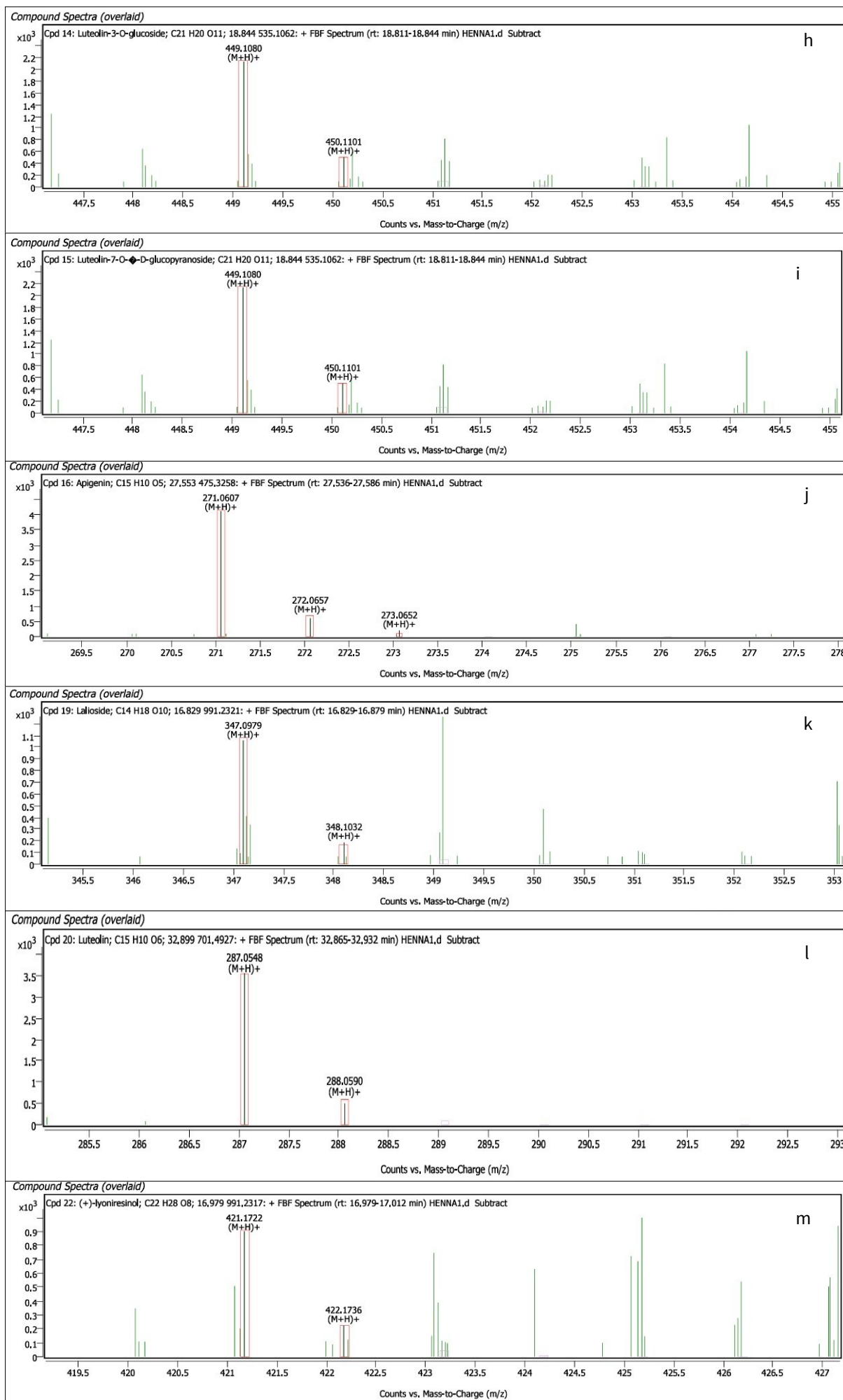
ADMET analysis

This analysis can help predict the performance of a drug in the body (42,43), which is done by ADMET analysis.

Molecular dynamic simulations

AutoDock Vina and Discovery Studio give a very stringent overview of the proteins-ligand interactions (44). This limitation can be overcome by molecular dynamic simulations between complexes, in a solvation box, for a specific time-period (45,46). From the previous analysis, lawsone and caffeic acids were shortlisted which were further analysed (47). GROMACS is a versatile program and can be easily customized. For charge neutralization, 2 sodium atoms were added to each of the systems, which further aided in the smooth running of the simulations (48).

The force field used in the dynamic studies were CHARMM36 (charmm36-jul2022) and the SPC water model. The energy minimization was done, which ensures structural stability. Once the system was energy minimized and evaluated as positive for potential energy (Fig. 3), we achieved equilibration for the same. The equilibration was done for temperature and pressure both (The system was equilibrated within 1 ns). This was achieved at temperature equilibration in 1000 ps [1 ns, maintaining an average temperature of 300.022 K]. Temperature equilibration was done to increase system stability before simulation (49). In addition, the system got pressure equilibrated within 1000 ps, maintaining an average pressure of -0.258134 bar. The above process of energy minimization and equilibration (temperature & pressure) ensures structural stability, needed for further processing of molecular dynamic simulation by GROMACS. The system was simulated for 100 ns and the trajectory was recorded every 10 ps.
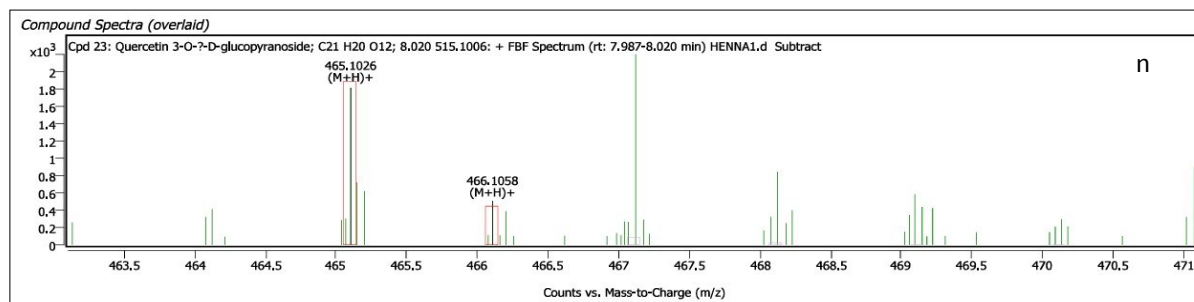
Compound Spectra (overlaid)

Cpd 1: Lawsone (2-Hydroxy-1,4-naphthoquinone); C10 H6 O3; 14.981 507.1108: + FBF Spectrum (rt: 14.981 min) HENNA1.d Subtract

a

Compound Spectra (overlaid)

Cpd 4: Chlorogenic acid; C16 H18 O9; 13.066 315.1056: + FBF Spectrum (rt: 13.049-13.099 min) HENNA1.d Subtract

b

Compound Spectra (overlaid)

Cpd 6: Caffeic acid; C9 H8O4; 28.452 449.1080: + FBF Spectrum (rt: 28.436-28.452 min) HENNA1.d Subtract

c

Compound Spectra (overlaid)

Cpd 7: Umbelliferone; C9 H6 O3; 16.979 991.2316: + FBF Spectrum (rt: 16.979-16.996 min) HENNA1.d Subtract

d

Compound Spectra (overlaid)

Cpd 8: Rutin; C27 H30 O16; 27.536 475.3257: + FBF Spectrum (rt: 27.520-27.536 min) HENNA1.d Subtract

e

Compound Spectra (overlaid)

Cpd 10: Quercetin; C15 H10 O7; 22.641 563.2098: + FBF Spectrum (rt: 22.607-22.691 min) HENNA1.d Subtract

f

Compound Spectra (overlaid)

Cpd 11: Kaempferol; C15 H10 O6; 32.899 701.4927: + FBF Spectrum (rt: 32.865-32.932 min) HENNA1.d Subtract

g

Fig. 2. MS spectrum of phytochemicals.

a) Lawsone; b) Chlorogenic acid; c) Caffeic acid; d) Umbelliferone; e) Rutin; f) Quercetin; g) Kaempferol; h) Luteolin-3-O-glucoside; i) Luteolin-7-O-β-D-glucopyranoside; j) Apigenin; k) Lalioside; l) Luteolin; m) Lyoniresinol; n) Quercetin-3-O-β-D-glucopyranoside

Table 1. Identified phytochemicals with their PubChem ID and formula

| Sl. No	Name	PubChem ID	Chemical formula
1	Lawsone	6755	$C_{10}H_6O_3$
2	Chlorogenic acid	1794427	$C_{16}H_{18}O_9$
3	Caffeic acid	689043	$C_9H_8O_4$
4	Umbelliferone	5281426	$C_9H_6O_3$
5	Rutin	5280805	$C_{27}H_{30}O_{16}$
6	Quercetin	5280343	$C_{15}H_{10}O_7$
7	Kaempferol	5280863	$C_{15}H_{10}O_6$
8	Luteolin-3-O-glucoside	12309350	$C_{21}H_{20}O_{11}$
9	Luteolin-7-O-β-D-glucopyranoside	5280637	$C_{21}H_{20}O_{11}$
10	Apigenin	5280443	$C_{15}H_{10}O_5$
11	Lalioside	189452	$C_{14}H_{18}O_{10}$
12	Luteolin	5280445	$C_{15}H_{10}O_6$
13	Lyoniresinol	11711453	$C_{22}H_{28}O_8$
14	Quercetin-3-O-β-D-glucopyranoside	12304324	$C_{21}H_{20}O_{12}$
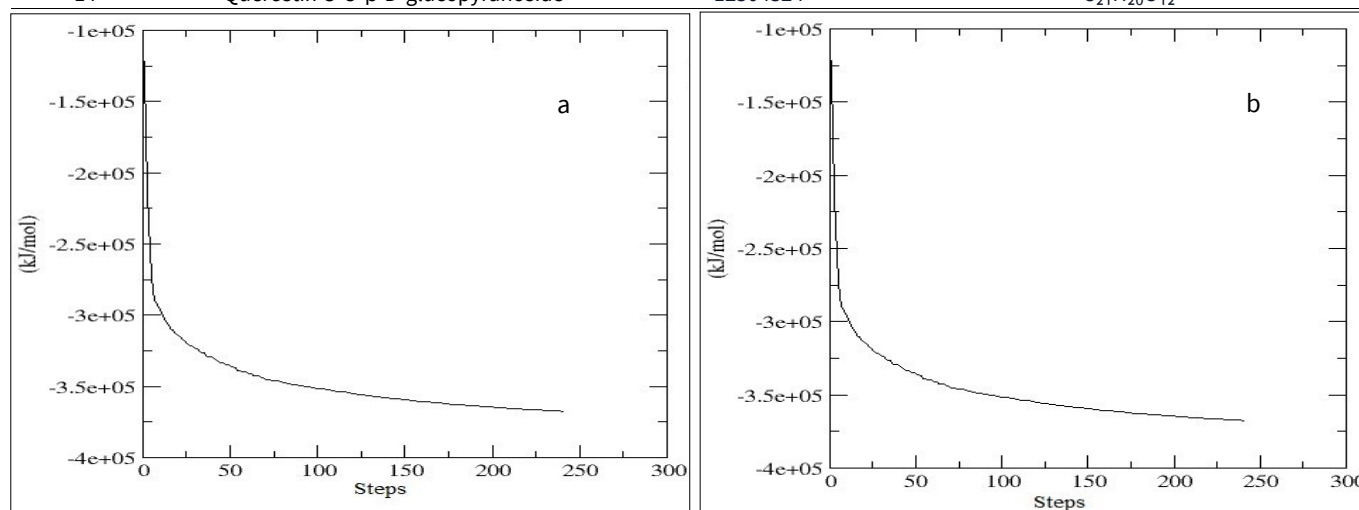
Fig. 3. Potential energy plots.

a) Lawsone; b) Caffeic acid complex with Nsp3 protein

Molecular Mechanics Poisson-Boltzmann Surface Area (MMPBSA) was employed to calculate the free energy of binding (△G) of complexes using g_mmpbsa tool of GROMACS (50). This can be calculated as:

$$\triangle G_{bind} = G_{complex} - (G_{protein}) + Gl_{ligand}$$

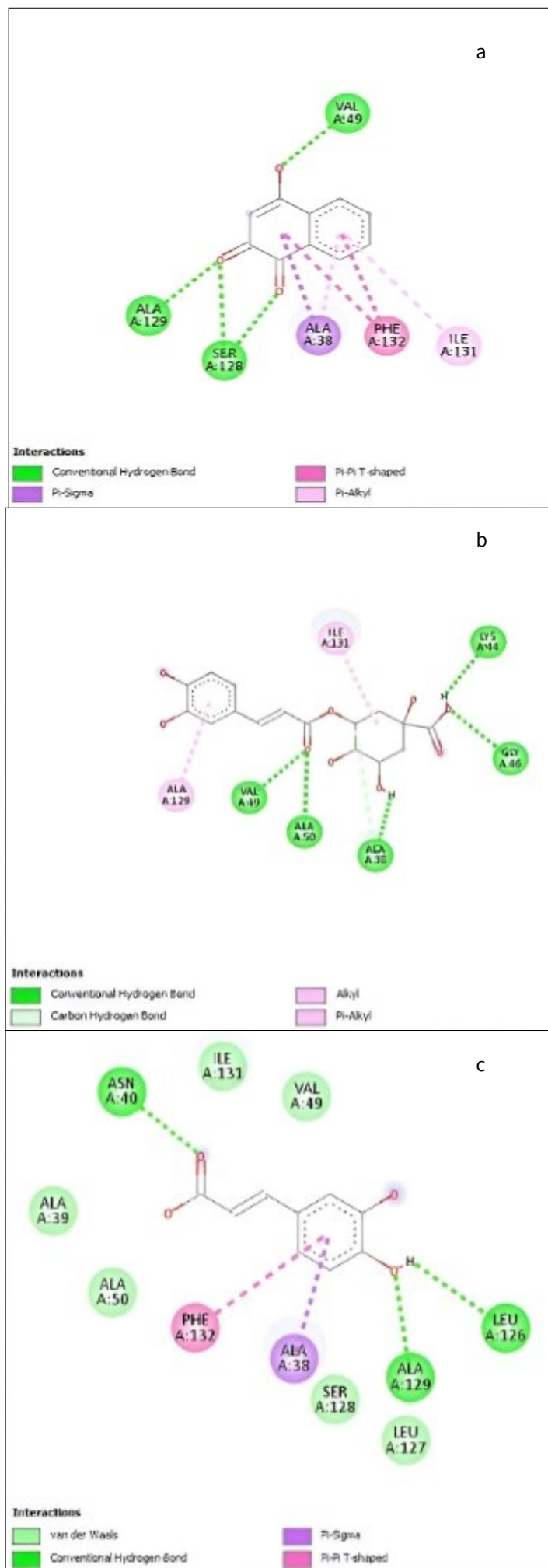## Results and Discussion

Docking analysis of phytochemicals

For a good drug candidate, strong interactions with the protein are required (Fig. 4, Table 2) Negative energy values define a spontaneous interaction. More negative values indicate a strong binding affinity between protein-ligand complexes (40,51). The highest number of hydrogen bonds was shown by Luteolin-3-O-glucoside. Lawsone, was also seen to exhibit a high binding affinity of -7.4 kcal/mol.
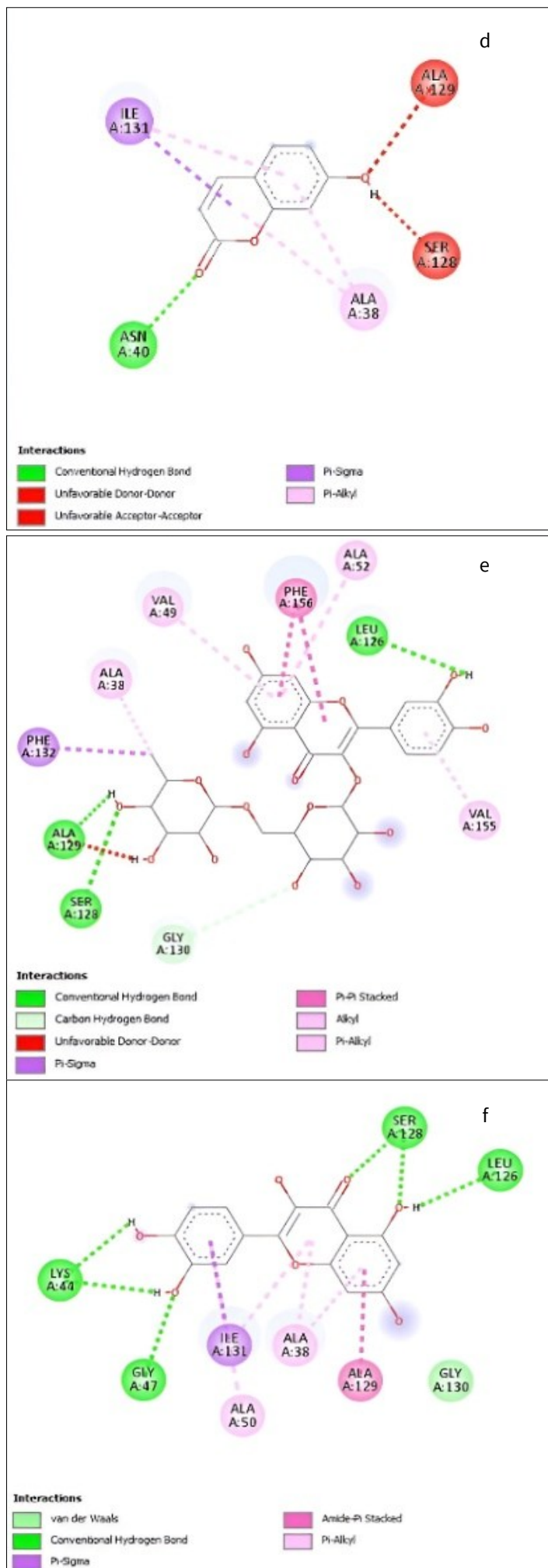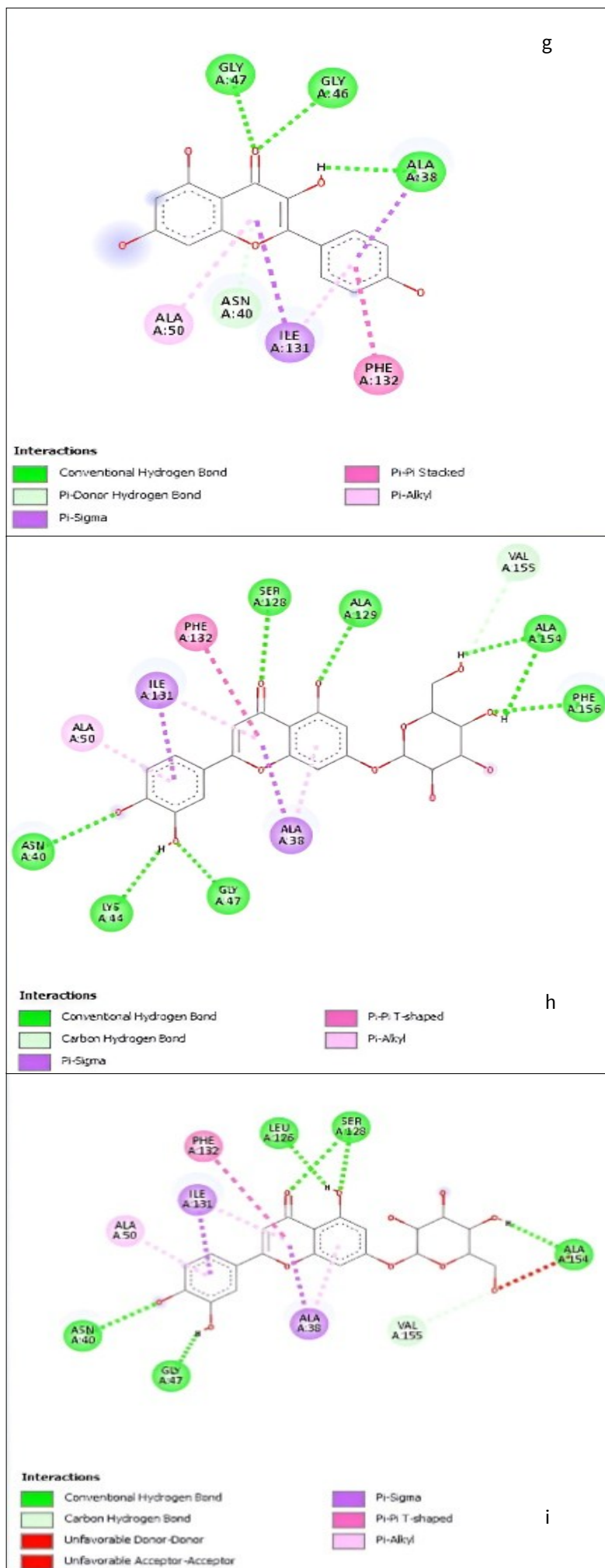
Phytochemical screening

The drug candidacy can be very easily comprehended by Lipinski's rule of 5, where lipophilicity (logP) of less than 5, less than 10 hydrogen bond acceptors, less than 5 hydrogen bond donors and a molecular mass of less than 500 Dalton, will increase the drug candidacy (Table 3). All the phytochemicals showed good performance, except chlorogenic acid, luteolin-3-O-glucoside, luteolin-7-O-β-D-glucopyranoside, quercetin-3-O-β-D-glucopyranoside and lalioside.
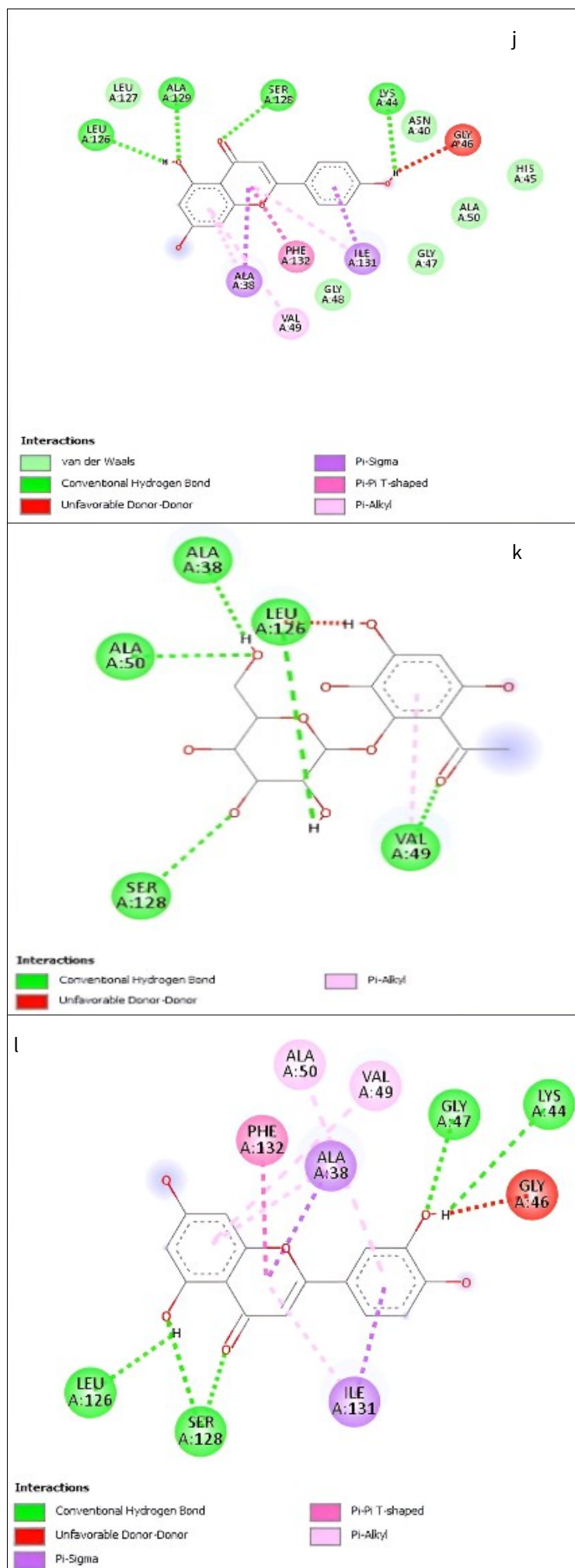
In silico ADME analysis

Gastrointestinal absorption was good for all the phytochemicals except, rutin, luteolin-3-O-glucoside, chlorogenic acid and luteolin-7-O-β-D-glucopyranoside (Table 4). A score of >0.55 is considered good. Lawsone shows the highest bioactivity score, of 0.85, which is an indicator of the extent and rate at which the

d

Interactions

■ Conventional Hydrogen Bond   ■ Pi-Sigma
■ Unfavorable Donor-Donor   ■ Pi-Alkyl
■ Unfavorable Acceptor-Acceptor

e

Interactions

■ Conventional Hydrogen Bond   ■ Pi-Pi Stacked
■ Carbon Hydrogen Bond   ■ Alkyl
■ Unfavorable Donor-Donor   ■ Pi-Alkyl
■ Pi-Sigma

f

Interactions

■ van der Waals   ■ Amide-Pi Stacked
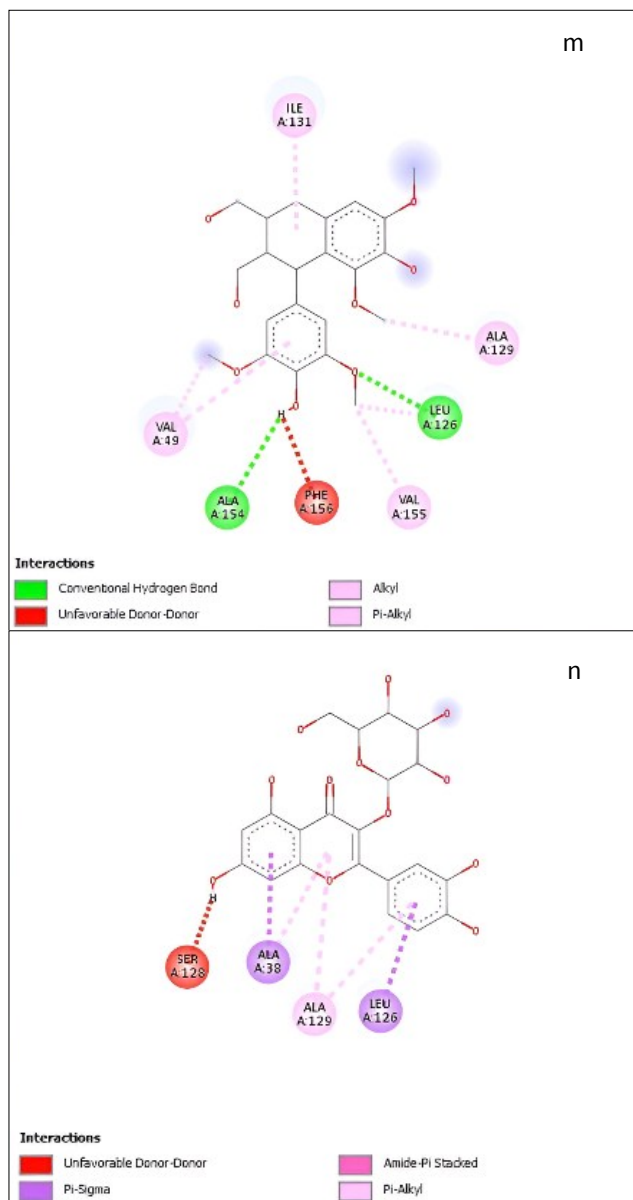■ Conventional Hydrogen Bond   ■ Pi-Alkyl
■ Pi-Sigma

Fig. 4. Binding residues of protein Nsp3 with phytochemicals.

a) Lawsone; b) Chlorogenic acid; c) Caffeic acid; d) Umbelliferone; e) Rutin; f) Quercetin; g) Kaempferol; h) Luteolin-3-O-glucoside; i) Luteolin-7-O-β-D-glucopyranoside; j) Apigenin; k) Lalioside; l) Luteolin; m) Lyoniresinol; n) Quercetin-3-O-β-D-glucopyranoside

Table 2. Binding residues of phytochemicals with protein Nsp3

Sr. No	Name	Binding affinity (kcal/mol)	H bonds	Binding residues
1.	Lawsone	-7.4	4	Ala 129 (A), Ser 128 (A), Val 49 (A)
2.	Chlorogenic acid	-8.2	5	Val 49 (A), Ala 50 (A), Ala 38 (A), Gly 46 (A), Lys 44 (A)
3.	Caffeic acid	-7.0	3	Ala 129 (A), Asn 40 (A), Leu 126 (A)
4.	Umbelliferone	-6.7	1	Asn 40 (A)
5.	Rutin	-9.0	3	Leu 126 (A), Ala 129 (A), Ser 128 (A)
6.	Quercetin	-8.4	6	Ser 128 (A), Leu 126 (A), Lys 44 (A), Gly 47 (A)
7.	Kaempferol	-8.2	3	Gly 46 (A), Gly 47 (A), Ala 38 (A)
8.	Luteolin-3-O-glucoside	-10.0	8	Ser 128 (A), Ala 129 (A), Ala 154 (A), Phe 156 (A), Gly 47 (A), Lys 44 (A), Asn 40 (A)
9.	Luteolin-7-O-β-D-glucopyranoside	-10.0	6	Leu 126 (A), Ser 128 (A), Ala 154 (A), Gly 47 (A), Asn 40 (A)
10.	Apigenin	-8.7	4	Leu 126 (A), Ala 129 (A), Ser 128 (A), Lys 44 (A),
11.	Lalioside	-7.5	5	Ala 38 (A), Ala 50 (A), Ser 128 (A), Leu 126 (A), Val 49 (A),
12.	Luteolin	-8.9	5	Leu 126 (A), Gly 47 (A), Ser 128 (A), Lys 44 (A),
13.	Lyoniresinol	-6.5	2	Ala 154 (A), Leu 126 (A)
14.	Quercetin-3-O-β-D-glucopyranoside	-7.9	-	-

Table 3. Summary of ligands and their characters and violations to Lipinski's rule of 5

SL. NO	Compound Name	Molecular Mass (g/mol)	H bond acceptors	H bond donors	LOG P	Molar refractivity	Violations
1	Lawsone	174.15	3	1	0.83	46.39	0
2	Chlorogenic acid	354.31	9	6	0.96	83.50	1
3	Caffeic acid	180.16	4	3	0.97	47.16	0
4	Umbelliferone	162.14	3	1	1.44	44.51	0
5	Rutin	610.52	16	10	1.58	141.38	3
6	Quercetin	302.24	7	5	1.63	78.03	0
7	Kaempferol	286.24	6	4	1.70	76.01	0
8	Luteolin-3-O-glucoside	448.38	11	7	0.72	108.13	2
9	Luteolin-7-O-β-D-glucopyranoside	448.38	11	7	1.83	108.13	2
10	Apigenin	270.24	5	3	1.89	73.99	0
11	Lalioside	346.29	10	7	0.75	76.85	1
12	Luteolin	286.24	6	4	1.85	76.01	0
13	Lyoniresinol	420.45	8	4	2.92	110.31	0
14	Quercetin-3-O-β-D-glucopyranoside	464.38	12	8	2.11	110.16	2

Table 4. Analysis of phytochemicals using Swiss ADME analysis

Name	LOG S	GIA	BBB	P-gp	CYP3A4 inhibitor	CYP1A2 inhibitor	Bioavailability score
Lawsone	-1.80	High	Yes	No	No	No	0.85
Chlorogenic acid	-1.62	Low	No	No	No	No	0.11
Caffeic acid	-1.89	High	No	No	No	No	0.56
Umbelliferone	-2.46	High	Yes	No	No	Yes	0.55
Rutin	-3.30	Low	No	Yes	No	No	0.17
Quercetin	-3.16	High	No	No	Yes	Yes	0.55
Kaempferol	-3.31	High	No	No	Yes	Yes	0.55
Luteolin-3-O-glucoside	-3.65	Low	No	No	No	No	0.17
Luteolin-7-O-β-D-glucopyranoside	-3.65	Low	No	Yes	No	No	0.17
Apigenin	-3.94	High	No	No	Yes	Yes	0.55
Lalioside	-1.23	Low	No	No	No	No	0.55
Luteolin	-3.71	High	No	No	Yes	Yes	0.55
Lyoniresinol	-3.53	High	No	Yes	No	No	0.55
Quercetin-3-O-β-D-glucopyranoside	-3.04	Low	No	No	No	No	0.17

ligand enters the system, accessing the site of action. Ligands that are p-glycoprotein substrates, can substantially reduce drug metabolism, hence are not acceptable. CYP enzymes are very important for drug absorption and metabolism, inhibitors of which, are discouraged. Quercetin, kaempferol, apigenin and luteolin don't satisfy the criteria, whereas lawsone and caffeic acid, can serve as prospective drug candidates that qualify the screening parameters successfully.

Bioactivity score

The phytochemicals showed a high to moderate bioactivity score, apart from umbelliferone, which showed lowest score in kinase Inhibition of -1.30. For G protein-coupled receptor (GPCR), Ion channel modulator (ICM), Enzyme inhibitor (EI), Protease inhibitor (PI) and Nuclear receptor ligand (NRL), high bioactivity score was shown by chlorogenic acid (Table 5).

AMDET analysis

All 14 phytochemicals were tested for ADME and toxicity to further screen the compound's candidature as effective drugs (Table 6). Regarding the absorption parameters, all phytochemicals showed optimal values. The distribution result showed that the compounds are well distributed in tissue. All compounds also showed a good renal elimination by not being renal organic cation transporter 2 (OCT2) substrates. AMES toxicity was not shown for the selected compound list. Only umbelliferone showed hepatotoxicity. Comparing all the parameters, lawsone and caffeic acid showed best results among the ligands.

Molecular Dynamic (MD) simulations

A dynamic interaction between the complexes can be understood by MD simulations. Docking methods give a stringent view, which can be easily overcome by simulation studies (45). These simulation studies also help in making the process of drug discovery less time

Table 5. Bioactivity score of the phytochemicals with the Molinspiration software

SL. NO	Name	GPCR	ICM	KI	NRL	PI	EI
1	Lawsone	-0.76	-0.13	-0.41	-0.86	-0.74	0.20
2	Chlorogenic acid	0.29	0.14	-0.00	0.74	0.27	0.62
3	Caffeic acid	-0.48	-0.23	-0.81	-0.10	-0.79	-0.09
4	Umbelliferone	-1.22	-0.72	-1.30	-0.92	-1.30	-0.35
5	Rutin	-0.05	-0.52	-0.14	-0.23	-0.07	0.12
6	Quercetin	0.06	-0.19	0.28	0.36	-0.25	0.28
7	Kaempferol	-0.10	-0.21	0.21	0.32	-0.27	0.26
8	Luteolin-3-O-glucoside	0.10	-0.01	0.18	0.28	-0.02	0.43
9	Luteolin-7-O-β-D-glucopyranoside	0.09	-0.02	0.15	0.27	-0.01	0.42
10	Apigenin	-0.07	-0.09	0.18	0.34	-0.25	0.26
11	Lalioside	0.03	0.06	-0.14	0.07	-0.01	0.38
12	Luteolin	-0.02	-0.07	0.26	0.39	-0.22	0.28
13	Lyoniresinol	0.07	0.00	-0.18	-0.05	-0.02	0.10
14	Quercetin-3-O-β-D-glucopyranoside	0.06	-0.04	0.13	0.20	-0.06	0.42

Table 6. ADMET prediction of the phytochemicals

Sl.NO	MOLECULE	Absorption			Distribution				Excretion		Toxicity							
		Caco₂	HIA	Skin	VDss	FU	BBB	CNS	TC	Renal OCT2	AMES	MTDD	hERG I	hERG II	LD50	HT	SS	MT
1.	Lawsone	1.19	93.8	-3.04	0.01	0.01	-0.23	-2.79	0.15	No	No	0.97	No	No	1.82	No	No	1.57
2	Chlorogenic acid	-0.84	36.4	-2.74	0.58	0.66	-1.41	-3.86	0.30	No	No	-0.13	No	No	1.97	No	No	5.74
3	Caffeic acid	0.63	69.4	-2.72	-1.09	0.53	-0.65	-2.61	0.50	NO	NO	1.14	No	No	2.38	No	No	2.24
4	Umbelliferone	1.21	94.5	-2.6	0.03	0.43	-0.28	-2.74	0.70	No	No	0.68	No	No	2.04	Yes	No	1.71
5	Rutin	-0.95	23.4	-2.74	1.66	0.18	-1.89	-5.18	-0.37	NO	NO	0.45	No	Yes	2.49	No	No	7.67
6	Quercetin	-0.23	77.2	-2.74	1.56	0.20	-1.09	-3.06	0.49	NO	NO	0.49	No	No	2.47	No	No	3.72
7	Kaempferol	0.03	74.2	-2.74	1.27	0.17	-0.94	-2.23	0.47	No	No	0.53	No	No	2.44	No	No	2.88
8	Luteolin-3-O-glucoside	0.28	30.3	-2.74	1.35	1.35	-1.63	-3.91	0.47	No	No	0.58	No	No	2.54	No	No	5.94
9	Luteolin-7-O-β-D glucopyranoside	0.24	37.5	-2.73	0.88	0.22	-1.56	-3.93	0.47	No	No	0.58	No	No	2.54	No	No	6.34
10	Apigenin	1.00	93.2	-2.73	0.82	0.15	-0.73	-2.06	0.56	No	No	0.32	No	No	2.45	No	No	2.43
11	Lalioside	-0.86	24.2	-2.73	0.80	0.71	-1.49	-4.14	0.52	No	No	0.24	No	No	2.40	No	No	6.88
12	Luteolin	0.09	81.1	-2.73	1.15	0.17	-0.91	-2.25	0.49	No	No	0.49	No	No	2.45	No	No	3.16
13	Lyoniresinol	-0.27	72.2	-2.73	0.19	0.05	-1.11	-1.11	0.37	No	No	0.41	No	Yes	2.08	No	No	2.46
14.	Quercetin-3-O-β-D-glucopyranoside	0.24	47.9	-2.73	1.84	0.23	-1.68	-4.04	0.39	No	No	0.56	No	Yes	2.54	No	No	8.06

consuming for the industry (52). We performed simulations for the best ligand-protein complexes, as was evident by previous screening analysis, (lawsone-Nsp3 and caffeic acid-Nsp3), for 100 ns, to finally select the best candidate.

Root Mean Square Deviation (RMSD)

We can say that Lawsone-Nsp3 and Caffeic acid-Nsp3 show very low RMSD, within 0.1-0.15 nm, (expected and stable range is 0.1-0.7nm with respect to size of the protein) till 100 ns. RMSD values also show that the variation is insignificant in terms of its structure which indicates a high stability (53) (Fig. 5). For the excellent category, RMSD values need to be less than 0.05 nm (54). Hence, lawsone and caffeic acid can serve as potential good hits based on their very low RMSD values.

Root Mean Square Fluctuation (RMSF)

From the results, the RMSF observed range is less than 0.3 nm for caffeic acid and lawsone complexes, indicating no significant change in the structure of protein-ligand complexes, suggesting strong and compact complexes (Fig. 6). Glycine and aspartic acid residues show highest mobility, contributing to protein flexibility.

Radius of gyration radius of gyration (Rg)

The radius of gyration measures protein and ligand compactness, throwing light on the folding properties of the protein (55,56). Small values indicate a compact complex. High Rg values indicate unstable protein folding and complexes (57). A stable value of the radius of gyration indicates stable protein folding. Analysis shows that the radius of gyration is relatively stable for both the Lawsone-Nsp3 and Caffeic acid-Nsp3 complexes. The stable graph (black colour) shows that the protein can maintain its compactness (Fig. 7).

Thermodynamic properties of the system

To calculate the free energy of binding ($\triangle$G) of complexes, MMPBSA was employed (Table 7). MMPBSA calculated the various forces (Table 7) for both complexes. Both complexes remain stable for 100 ns. It was found that the binding energy for 'caffeic acid' is -16.32 kcal/mol, while -20.72 kcal/mol for the 'lawsone'. These energies represent stronger binding for lawsone. In addition, energy less than -7 kcal/mol shows stronger binding. The complex 'Lawsone' is more stable than the 'Caffeic acid' based on lower free energy.

The medicinal importance of *Lawsonia inermis* is well established, given the rich secondary metabolites (9). When it comes to plant-based phytochemicals, we usually see flavonoids as effective antiviral agents, like diosmin (58–60). This analysis, however, show that, between caffeic acid and lawsone, a naphthoquinone, which is also the main pigment, emerging as a potential antiviral candidate (61,62). Naphthoquinones are derivatives of the shikimic acid pathway and have well established medicinal benefits (63,64). Earlier studies reveal the effective role of liposome-encapsulated derivatives of naphthoquinones in inhibiting HSV infection (65). Even novel derivatives of 1,4-naphthoquinones have shown effective antimicrobial functions (66). 3,5,8-trihydroxy-6-methoxy-2-(5-oxohexa-1,3-dienyl)-naphthanthene-1,4-dione, a naphthoquinone produced by a fungus was reported to reduce inflammation and inhibit transcription (hindering the virus pathogenicity) in SARS-CoV-2 infected cells (67). From our study, it was interesting to see that, for the very first time, between caffeic acid and lawsone, lawsone yielded the best results, against Nsp3 protein. Naphthoquinones achieve the antiviral property mainly by inhibiting the Na⁺, K⁺-
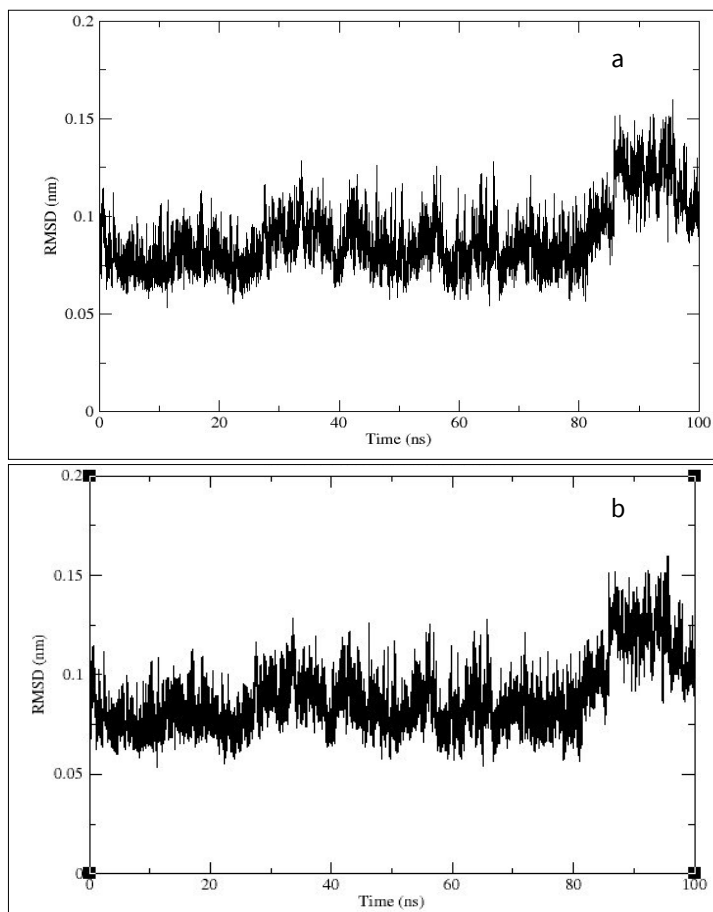
Fig. 5. RMSD plots.

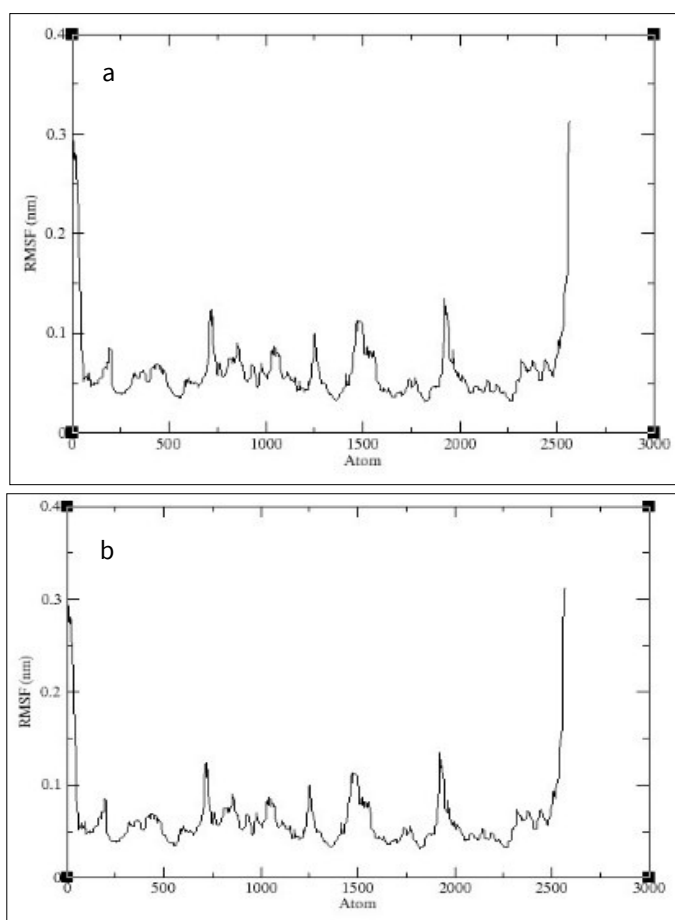a) Lawsone; b) caffeic acid complex with protein Nsp3

Fig. 6. RMSF plots.

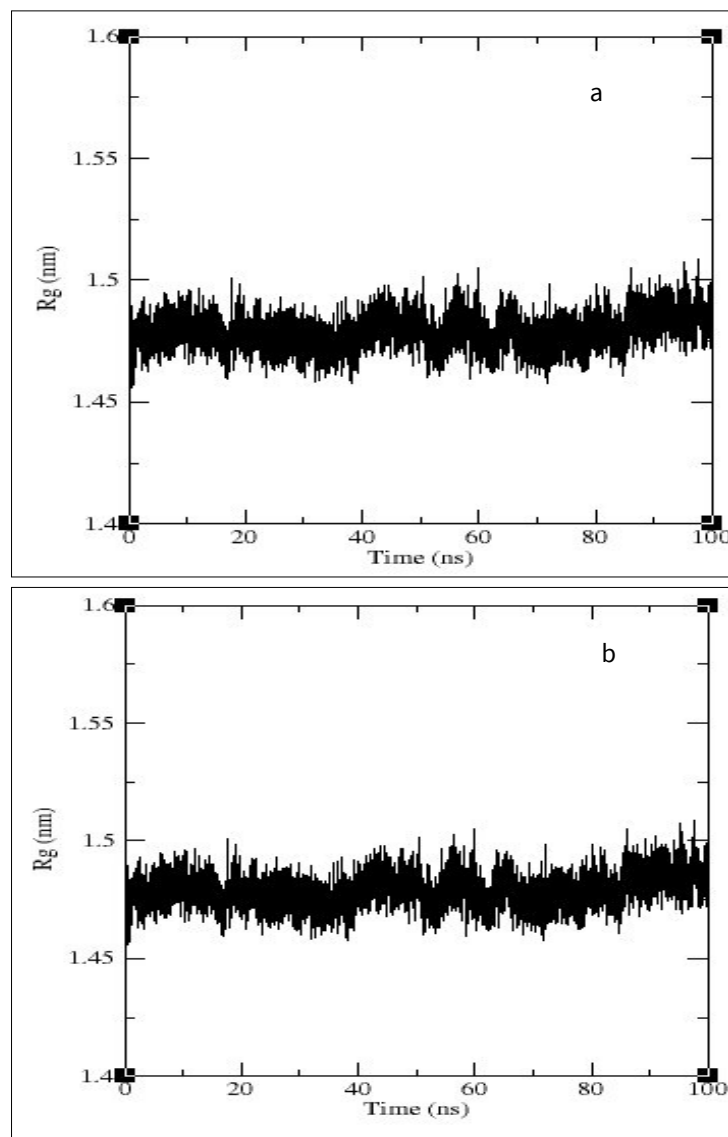a) Lawsone; b) caffeic acid complex with protein Nsp3

Fig. 7. Radius of gyration (Rg) plots.

a) Lawsone; b) caffeic acid complex with protein Nsp3

Table 7. Energies of the complexes

S.No	Complex	Vanderwaal energy (kcal/mol)	Electrostatic energy (kcal/mol)	Polar and Non Polar solvation energy (kcal/mol)	SASA energy (kcal/mol)	Free Binding energy (kcal/mol)
1	Lawsone-Nsp3	-1155.92	-11392.83	-1904.39	65.02	-20.72
2	Caffeic Acid-Nsp3	-1149.59	-11394.57	-1917.29	66.26	-16.32

ATPase and by interfering with the steps of attachment and penetration of the virus (65,68,69).

The importance of the Nsp3 protein lies in its many functions to ensure the survival of the virus, by promoting of replication, transcription and translation (26,27,29,32,70). A triterpene glycoside, suavissimoside, was reported to hinder the activity SARS-CoV M^{Pro}, having a binding energy of -8.19 kcal/mol (71). Sulfuretin, was also reported to show good binding results with Nsp3 (72). Chlorogenic acid from *Lawsonia inermis* also showed effective binding and interaction with NS1 of Dengue virus (15), with a binding energy of -43.39 kJ/mol. In this study, we see the main pigment of the plant, lawsone, to be effective against Nsp3. No previous studies were found, that explored the interactions of lawsone against Nsp3, proving the uniqueness of the analysis. Lawsone and caffeic acid satisfied most of the screening tests, which were further simulated for a time of 100 ns, by MD simulations. Analysis of RMSD, RMSF and Rg showed the ligand-protein complexes, analysing which, concluded strong and compact complexes. Binding free energy for the complex lawsone was -20.72 kcal/mol, suggesting favourable energy parameters as compared to caffeic acid complex. From the results it is clear that the Lawsone-Nsp3 complex is more stable. This analysis shows better binding affinity as compared to suavissimoside to SARS-CoV M^{Pro}, with an affinity of -8.19 kcal/mol (71). This was similar to the energy of a few best docked ligands against Nsp3, showing an affinity of more than -8.5 kcal/mol (73). When comparing with previous works (74,75) ligand binding affinity was seen to be higher for our analysis, where the ligands interacted with the inhibitory binding residues of Nsp3 protein.

Our analysis yielded better results, showing the possibility of a natural dying agent, as a potential antiviral compound. This analysis also highlights the interactions of lawsone, a naphthoquinone, against Nsp3 protein. In hindsight, this work can successfully establish the plant in the medicinal sector.

Conclusion

We can conclude from the analysis that a naphthoquinone, lawsone, which is the main pigment of the plant, can serve as a potential candidate against Nsp3 protein of SARS-CoV2, providing a plant-based alternative that can successfully bind and inhibit the protein, needed for protein processing by the virus. Further isolation of lawsone and application in *in vitro* methods, can strengthen the candidature of the ligand against the protein's functions.

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Authors' contributions

DD and DS wrote paper, collected data and analysed data. DR, SD and BM designed research and edited the manuscript. All authors read and approved the final manuscript.

Compliance with ethical standards

Conflict of interest: Authors do not have any conflict of interests to declare.

Ethical issues: None

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Additional information