



RESEARCH ARTICLE

Kigelia africana (Lam.) Benth. fruit inhibits iron-induced lipid peroxidation and α -amylase enzyme activity

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Abstract

Insufficiencies in insulin secretion and/or action cause diabetes mellitus (DM), a complex condition characterized by abnormal blood glucose levels. Diabetes is often treated by modifying either glucose metabolism, lipids metabolism, or both. A direct correlation has been found between high levels of lipid peroxide in individuals with diabetes and lower intracellular antioxidant activity. Therefore, this study was designed to investigate the effect of the aqueous extract of *Kigelia africana* fruit on the carbohydrate hydrolyzing enzyme, α -amylase, and FeSO₄-induced lipid peroxidation in the rat pancreas. The *in vitro* antioxidant potential, including assays for (1,1 diphenyl-2-picryl hydrazyl (DPPH), ferric reducing antioxidant property (FRAP), nitric oxide (NO), hydroxyl (OH) scavenging, and Fe²⁺ chelating abilities, of the fruit extract was also evaluated. The effect of the fruit extract on the α -amylase enzyme, and FeSO₄-induced lipid peroxidation in the rat pancreas was also evaluated. *K. africana* at a concentration of 0.5 mg/mL and 1 mg/mL scavenged DPPH radicals, with the lower concentration showing significant differences compared to the control. The *K. africana* fruit extract reduced ferric compounds significantly compared to the standard. The OH* radical scavenging results revealed that all concentrations of *K. africana* fruit scavenged OH* radicals without significant differences compared to the control. *K. africana* fruit also chelated iron significantly compared to the control. The fruit extract inhibited iron-induced lipid peroxidation in the rat pancreas and α -amylase activity compared to the control. *K. africana* fruit displayed promising potential in inhibiting α -amylase activity and associated lipid peroxidation in diabetics. Therefore, it holds strong promise for use in diabetes treatment, validating its traditional use for managing the condition.

Keywords

Diabetes; Malondialdehyde; ferric reducing antioxidant property; Fe²⁺ chelating abilities; *Kigelia africana* fruit

Introduction

Insufficiencies in insulin secretion and/or action cause diabetes mellitus (DM), a complex condition characterized by abnormal blood glucose levels. Diabetes is often treated by modifying either glucose metabolism, lipids

metabolism, or both (1). Millions of people worldwide suffer from diabetic complications, with nearly 10% of the global population affected by this serious endocrine condition (2). By 2030, it is projected that there will be 439 million people with diabetes worldwide (3). Diabetes is associated with the growth of the contemporary lifestyle globally (4). Our society experiences substantial morbidity and mortality due to DM-related vascular impairment (5–7). Blood vessels and nerve tissues are predominantly damaged as a result of this metabolic imbalance, leading to significant cellular and organ damage (8).

Lipid peroxidation is regarded as the primary mechanism responsible for the toxic processes that lead to the oxidative destruction of cell components. Therefore, lipid peroxidation plays a significant role in human health and disease (9–12). Scientific evidence connects poor glycemic control, increased production of lipid oxidation products, and worsening metabolic health in diabetic individuals (13, 14). A direct correlation has been found between high levels of lipid peroxide in diabetes individuals and lower intracellular antioxidant activity (15). Therefore, the use of antioxidants may be useful in managing this condition (16–18). The importance of plant extracts in the management of diabetes is well known, and many medicinal plants have been identified for their usefulness in this regard (19). In hyperglycemic conditions, various factors, including impairment of the insulin mechanism, hyperlipidemia, and oxidative stress, contribute to secondary complications such as diabetic retinopathy, neuropathy, and cardiovascular diseases (20). Phenolic compounds such as flavonoids, phenolic acids, lignans, and stilbenes, found in plant-based foods, may be effective as nutraceuticals and supplemental treatments for diabetes (21) and its complications (22).

Kigelia africana (Lam.) Benth. is native to Africa and is most frequently found in the continent's southern, central, and western regions (23, 24). It is widely used to treat a variety of ailments and serves as an agroforestry tree. Due to its peculiar fruit and stunning deep red flowers, the plant was introduced to several countries in Southeast Asia, including India, Pakistan, China, the Philippines, and Iraq. Today, it is mainly grown as a decorative tree and is commonly found in gardens and parks (25). Traditional African healers have used formulations for various parts of *K. africana* to treat a variety of skin issues, as well as diarrhea, rheumatism, constipation, ulcers, wounds, gonorrhea, cancer, abscesses, and many other conditions (26). This plant is rich in terpenes, terpenoids, and flavonoids. According to research, *K. africana* and its contents have antioxidant potential, anti-inflammatory, anti-cancer, and antidiabetic properties (27). Therefore, it is imperative to investigate the α -amylase enzyme activity and lipid peroxidation in the pancreas to determine the mechanism of action of *K. africana* in the treatment of diabetes mellitus. Hence, the aim of this study was to determine the effect of *K. africana* fruit on the α -amylase enzyme and iron-induced lipid peroxidation in the rat pancreas.

Materials and Methods

Plant Collection, Identification and Authentication

The *Kigelia africana* fruits were purchased from the Oja Oba market in Osogbo, Osun state, Nigeria. They were identified at the Bowen University Herbarium, Iwo, Osun state, and authenticated at the Forest Herbarium in Ibadan (FHI number: 111350). The fresh fruit was cut into small pieces and pulverized using an electric grinder. 1 kg of pulverised paste was extracted with 800 mL of distilled water for 72 hrs. The filtrate was concentrated using a freeze-drier to yield the aqueous extract of the plant. The dark-brown residue was stored in the refrigerator for further analysis.

Antioxidant Assays

Ferric reducing power was determined using Oyaizu's method (28), while the DPPH free radical scavenging ability was evaluated according to standard procedure (29). The extracts' ability to chelate Fe^{2+} was assessed using a modified Minotti and Aust technique (30), with a minor modification made by Oboh *et al.* (31). The Halliwell and Gutteridge method (32) was used to test the extracts' capacity to inhibit the $\text{Fe}^{2+}/\text{H}_2\text{O}_2$ -induced breakdown of deoxyribose.

α -amylase Inhibition Assay

To investigate the antidiabetic properties of *K. africana* fruit aqueous extract, we employed the α -amylase inhibitory protocol (33). The extract dilution (0–200 μL) was combined with 500 μL of 0.02 M sodium phosphate buffer (pH 6.9 with 0.006 M NaCl) containing 0.5 mg/mL of porcine pancreatic amylase and incubated for 10 min at 25 °C. Each tube was then filled with 500 μL of a 1% starch solution in 0.02 M sodium phosphate buffer (pH 6.9 with 0.006 M NaCl). The reaction mixture was incubated for 10 min at 25 °C before being stopped with 1.0 mL of dinitrosalicylic acid (DNSA). Subsequently, the mixture was cooled to room temperature after 5 min of incubation in a hot water bath. After diluting the reaction mixture with 10 mL of distilled water, the absorbance was measured at 540 nm. The α -amylase inhibitory activity was expressed as percentage inhibition.

Lipid Peroxidation Assay

Animals

10 male albino rats, weighing between 150 and 200 g, were used. Commercial farmers' mash pellets were used to feed the rats. They were kept at room temperature and fed ad libitum, except for the final 12 to 15 hrs before the experiment's end. The animals were housed in plastic cages at room temperature with a 12-hr light/12-hr dark cycle. All relevant national, institutional, and international regulations for the handling and use of animals were adhered to. Additionally, the experiment was approved by the institutional animal ethics committee at Bowen University, Iwo (BUI/BCH/2023/0001).

Preparation of Pancreas Homogenate

The pancreases were quickly extracted after euthanizing the rats using moderate diethyl ether anesthesia. The tissue was then homogenized in cold saline (1/5 w/v) using

ten up-and-down strokes in a Teflon glass homogenizer at a speed of roughly 1200 rev/min. A low-speed supernatant (S1) was retained for the lipid peroxidation experiment after centrifuging the homogenate for 10 min at 3000 × g (34).

Lipid Peroxidation and Thiobarbituric Acid Reactions

The modified protocol of Ohkawa *et al.* (35) was used for the lipid peroxidation assay. Briefly, 30 µL of 0.1 M pH 7.4 Tris-HCl buffer, 30 µL of freshly made 250 µM FeSO₄, and 100 µL of the S1 fraction were combined to form the reaction mixture (the operation was also carried out with 15 mM quercetin). After incubating the volume at 37 °C for 1 hr, 300 µL of water was added. Following the addition of 300 µL of 8.1% sodium dodecyl sulfate to the reaction mixture containing S1, 600 µL of acetic acid/HCl (pH 3.4) mixture, and 600 µL of 0.8% thiobarbituric acid were added to create the colour reaction. This mixture was then incubated at 100 °C for an hr. The absorbance was measured at 532 nm. The standard curve of MDA (Malondialdehyde) was used.

Statistical Analysis

Data (n=3) were presented as the mean ± standard deviation (SD) of three experiments. The data were analysed using One Way Analysis of Variance (ANOVA). A p-value less than 0.05 was considered statistically significant.

Results and Discussion

In vitro Antioxidant Activity

Free radical damage is etiologically connected to many chronic health problems. Antioxidants lessen the possibility of tissue damage brought on by free radicals by restricting the production of radicals, scavenging them, or encouraging their breakdown (36–40). Oxidative stress exacerbates a wide range of human disorders, such as diabetes, atherosclerosis, inflammatory arthritis, and cancer. The use of synthetic antioxidants has been limited due to their low solubility and possible health hazards

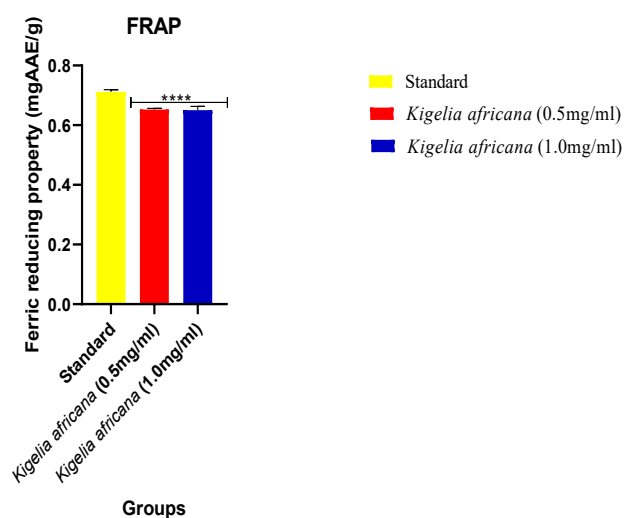


Fig. 1. Ferric reducing antioxidant property of aqueous extract of *Kigelia africana* fruit. Data are presented as mean ± SD (n=3). ****Values are statistically different at $p < 0.0001$ vs Standard. Standard = Ascorbic acid.

(41–44). In the recent quest for novel antioxidants, many plant species have been investigated (41, 45, 46). Less hazardous phyto-antioxidants, which are important medicinal components of food, have successfully decreased the susceptibility to reactive oxygen species (ROS) (47). Fig. 1 displayed the result of the ferric reducing antioxidant property (FRAP) of *Kigelia africana* fruit, demonstrating that *K. africana* at both concentrations (0.5 and 1 mg/mL) significantly reduced ferric radicals compared to the standard. The results of the ability to scavenge DPPH free radicals are depicted in Fig. 2, with *K. africana* (0.5 and 1 mg/mL) exhibiting the highest DPPH radical scavenging ability, statistically different from the control. Fig. 3 illustrates the ability of *K. africana* to chelate iron. At both concentrations of 0.5 and 1 mg/mL, significant differences were observed compared to the control. Fig. 4 presents *K. africana*'s capacity to scavenge hydroxyl radicals (OH), showing no discernible difference between its capacity and that of the control. *K. africana* is a substantial source of compounds that may be useful in reducing oxidative stress (48). Therefore, consuming medicinal plants with antioxidant potential appears to be

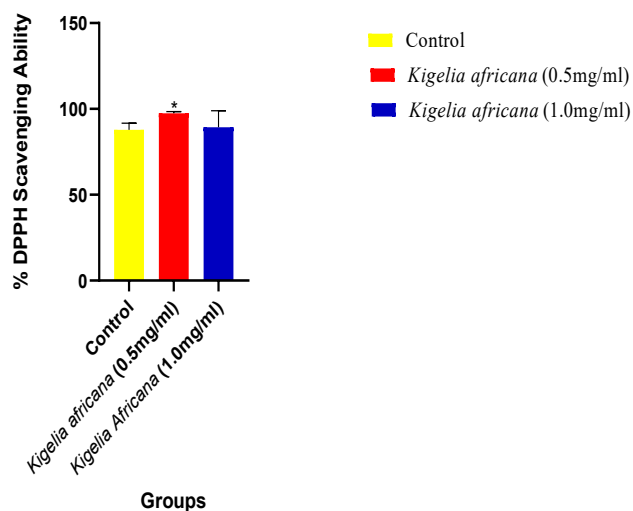


Fig. 2. DPPH scavenging ability of *Kigelia africana* fruit aqueous extract. Data are presented as mean ± SD (n=3). Values are statistically different at * $p < 0.05$ vs Control. Control = Quercetin.

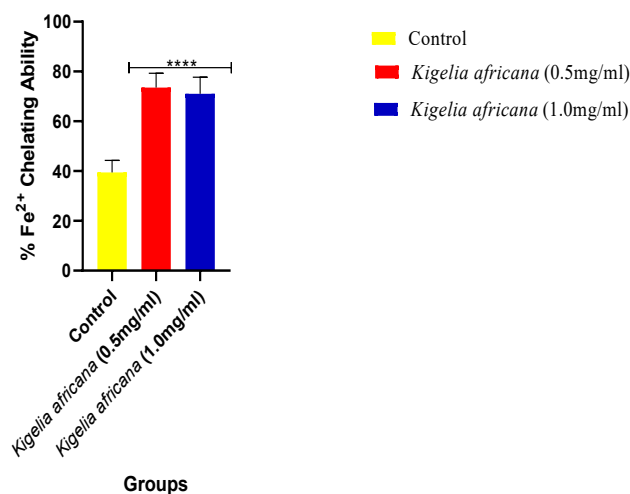


Fig. 3. Fe²⁺ chelating ability of *Kigelia africana* fruit aqueous extract. Data are presented as Mean ± SEM (n=3). Values are statistically different at **** $p < 0.0001$ vs Control. Control = Quercetin.

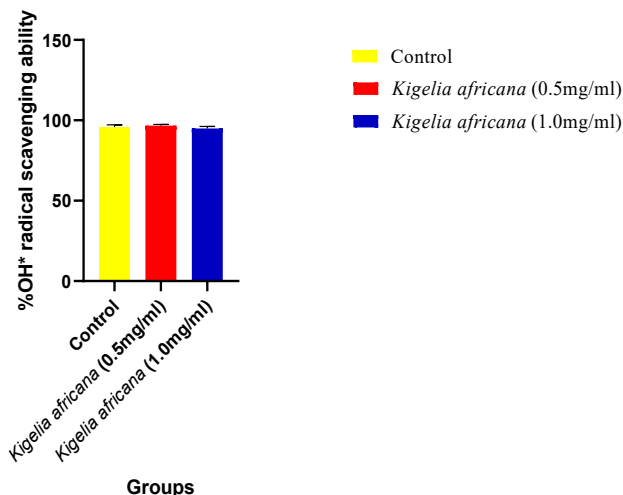


Fig. 4. OH[•] scavenging ability of *Kigelia africana* fruit aqueous extract. Data are presented as Mean \pm SEM (n=3). Control = Quercetin.

a workable substitute for natural antioxidant defense systems (49–51). Phenolics, which have a -OH functional group, are common components of medicinal plants and are used for several therapeutic applications, including the inhibition of free radicals. There is a comparable link between phenolics and antioxidant capacity (52–54). The high scavenging properties of the *K. africana* fruit aqueous extract may be due to the hydroxyl functional groups in the phenolic compound structure's skeleton.

α -amylase Inhibitory and Lipid Peroxidation Assays

While several chemicals with antioxidant activity have been identified in *K. africana* fruit (27, 49), little is known about how effectively these compounds function as antidiabetics (50, 51, 55–57). Fig. 5 presents the inhibition of α -amylase by *K. africana*, revealing that the concentrations of *K. africana* extract (0.5 and 1 mg/mL) did not significantly differ from the control in their ability to inhibit α -amylase activity. Fig. 6 depicts the inhibition of lipid peroxidation caused by iron sulfate (FeSO_4) in the rat pancreas by *K. africana* fruit aqueous extract. Both concentrations of *K. africana* extract (0.5 and 1 mg/mL) inhibited the production of MDA with no appreciable difference compared to the control. The Malondialdehyde (MDA) content significantly increased when 5 mM FeSO_4 was added to the rat pancreas homogenate (58). *K. africana* extract concentrations (0.5 and 1 mg/mL) reduced MDA generation more effectively than the control. The results of the α -amylase inhibition by *K. africana* fruit aqueous extract showed that *K. africana* concentrations (0.5 and 1 mg/mL) suppressed α -amylase activity more effectively than the control. Numerous compounds from botanicals are beneficial in the management of diabetes (59–61), with several medicinal plants possessing significant antioxidant activity also reported to have antidiabetic effects (54, 62). According to reports, one crucial component of treating type II diabetes is controlling human pancreatic α -amylase (54). Therefore, one of the primary strategies for managing diabetes is to delay the breakdown of starch by inhibiting α -amylase. Pancreatic α -amylase inhibitors slow down the digestion of carbohydrates, lowering the postprandial levels of blood glucose and slowing the rate of glucose absorption

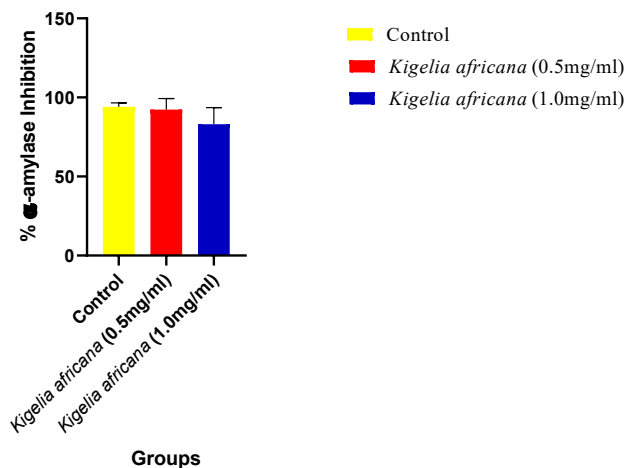


Fig. 5. α -amylase inhibitory effect of *Kigelia africana* fruit aqueous extract. Data are presented as Mean \pm SEM (n=3). Control = Arcabose.

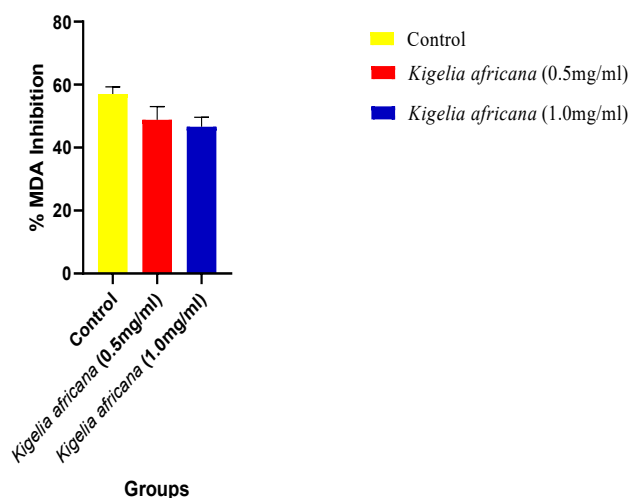


Fig. 6. Effect of *Kigelia africana* fruit aqueous extract on FeSO_4 -induced lipid peroxidation in rat pancreas. Data are presented as Mean \pm SEM (n=3). Control = Quercetin.

(63). The results of the current investigation also suggest one possible mechanism by which *K. africana* fruit might exhibit hypoglycemic effects: suppression of α -amylase activity, which would delay starch hydrolysis and subsequently lower postprandial hyperglycemia (PPHG) (64). The fruit of *K. africana* may possess anti-diabetic properties by preventing oxidation of pancreatic islet bio-membranes (65). However, the efficacy of *K. africana* fruit aqueous extract in alleviating diabetic conditions should not be solely based on α -amylase inhibitory activity. Therefore, *in vivo* assays and *in silico* studies need to be conducted to confirm this mechanism of action.

Conclusion

A *Kigelia africana* fruit exhibits promising potential to inhibit α -amylase activity and associated lipid peroxidation in diabetic individuals. Therefore, it holds strong promise for use in the treatment of diabetes, thereby validating its traditional medicinal use in managing the condition.

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Authors' contributions

EOF, ADO, and POA conceived the study and participated in its design and coordination. All authors carried out the study. POA performed the statistical analysis. All authors read and approved the final manuscript.

Compliance with ethical standards

Conflict of interest: Authors do not have any conflict of interests to declare.

Ethical issues: None.

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