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Research Article

Efficient synthesis of plant-mediated silver nanoparticles and their screening for antimicrobial activity

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Abstract

Now days, the development of safe, cost effective, reliable and eco-friendly processes for the synthesis of nanoparticles is an important aspect of nanotechnology. Among the various agents, plants show immense potential for the synthesis of nanoparticles. The bio-molecules found in plants induce reduction of Ag⁺ ions from silver nitrate to silver nanoparticles (AgNPs); therefore, in the present work, the aqueous leaves extract of the plant was used as reducing agent for the synthesis of silver nanoparticles. We synthesized extracellular silver nanoparticles using extract of the leaves of four different medicinal plants which act as a reducing agent at room temperature. The characteristic color change was observed on addition of plant extract to the silver nitrate solution due to their specific properties (Surface Plasmon Resonance). UV-Vis spectroscopy was used for the characterization of the silver nanoparticles. Green synthesized nanoparticles are evaluated for their antimicrobial activity against the Gram-positive and Gram-negative bacteria as well as two pathogenic fungi Aspergillus fumigatus and Curvularia lunata. The silver nanoparticles (SNPs) of selected plant parts have shown more toxicity towards bacterial species than that of the fungal species. Comparing with simple plant extracts, the SNPs exhibited greater antimicrobial efficacy and advantage over conventional antibiotics to which these microorganisms usually impart resistance.

Keywords

SNPs; Plant extract; Antimicrobial activity; Nanobiotechnology

Citation

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1 Introduction

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Nanotechnology refers to an extensive area of research with a unifying theme of controlling matter size from micrometer to nanometer which is known as nanoparticles, which constitute the fundamental building blocks of nanotechnology. their extensive applications, several artificial methods have been developed



for the synthesis of nanoparticles (1). Since metal nanoparticles show potent industrial applications, there is a growing interest in the biological and environmentally safe production of such particles. The artificial methods of nanoparticle synthesis include chemical and physical approaches, which are often expensive and potentially harmful to the environment. Recently, various biological methods such as use of microorganism (2,3), enzyme (4) and plant extract (5-7) have been increasingly considered as an alternative approach for the production of nanoparticles with specified properties.

Amongst these, plant extract offers several advantages over others as it is efficient, inexpensive, non-toxic and eco-friendly method of nanoparticles production (8). Previous studies revealed that the size, morphology, stability, and physicochemical properties of metal nanoparticles are strongly influenced by the experimental conditions (9-11). This property is predominantly significant for noble metals such as silver and gold, which have strong surface plasmon resonance oscillations (12). The antibacterial and antifungal properties of copper, silver, titanium and zinc efficiently control the growth of various microorganisms (13,14). A number of studies have shown that AgNPs prepared using plant extract exhibit higher antibacterial activity and strong toxicity against a wide range of microorganisms. Therefore, silver-based compounds have been used widely in many bactericidal applications. Excessive dosage and intensive use of antibiotics have brought the bacteria on the brink of increasing resistance against antibiotics, which necessitate an urgent need to develop efficient and antibiotics cost-effective approaches for development.

SNPs have been found to be a good substitute as these confers an inhibitory effect against many pathogens, particularly bacteria and fungi (15,16). The SNPs are increasingly being used in medicine to reduce infections in burn treatment and arthroplasty to prevent bacterial colonization on prostheses (17), catheters (18), vascular grafts, dental materials (19,20), stainless steel materials (21) and human skin (22). SNPs also exhibit a potent cytoprotective activity towards infected cells (23). In view of the importance of plant-mediated SNPs in medicine, the aqueous leaf extracts of three different medicinal plants i.e. Punica granatum, Aegle marmelos and Datura innoxia have been used as a reducing and stabilizing agent to synthesize silver nanoparticles. The silver nitrate was reduced to metallic silver on reaction with leaf extract of above mentioned medicinal plants, resulting in the synthesis of silver nanoparticles. The silver nanoparticles formed were characterized using ultravioletvisible spectroscopy. Antimicrobial assays using silver nanoparticles demonstrated activity against both the bacteria and fungi tested.

2 Materials and methods

2.1 Collection of plant materials and extract

The leaves of selected medicinal plants were collected from the experimental field of Guru Ghasidass University Bilaspur, Chattisgarh, India (Table 1). The leaves were thoroughly washed with distilled water to remove any traces of debris adhere to it and chopped into small pieces. The 5 gm of washed and chopped leaves are taken in a 250 ml Erlenmeyer flask with 100 ml of milli-Q water and this mixture was boiled for 15-20 min (24,25). The herbal aqueous extract was filtered in the separate conical flasks by standard filtration method using Whatman no. 1 filter paper and stored at 4°C. The leaf extract acts as a reducing agent for the synthesis of nanoparticles.

2.2 Synthesis and characterization of silver nanoparticles

different plant extracts of medicinal importance was used for the synthesis of silver nanoparticles. Silver nitrate (AgNO₃) solution was used as a source of metal for nanoparticle synthesis according to the process as given by Zargar et al (23). The 10 ml of filtered plant extract was added to 90 ml of aqueous solution of 1mM silver nitrate (AgNO₃) for the reduction of silver nitrate into Ag⁺ ions and kept for the incubation at room temperature. After 10-15 min, the color of the solution changed into yellowish orange and blackish, this indicates the formation of silver nanoparticles. The reduced silver nanoparticles solution was filtered through Whatman no.1 filter paper and the filtrate was qualitatively characterized by **UV-Visible** and measured (Spectrophotometer UV-vis 1800, Shimadzu, Kyoto, Japan).

2.3 UV-Visible spectra analysis

There are various techniques used for characterizing nanoparticles. However, nanoparticles characterization generally subject to their size, shape, surface area and dispersity. The preliminary and convenient way for nanoparticles characterization is the observation of color change with the addition of AgNO₃ solution. UV-vis spectral analysis was done by using UV-vis spectrophotometer at a resolution of 1nm in the range 900-1100 nm. The reduction of pure Ag⁺ ions to Ag⁰ was monitored by measuring the UVvis spectrum with the sampling of aliquots (0.1ml) of AgNPs solution and diluting the sample in 3ml distilled water. The de-ionized water was used for all UV-vis spectrum background correction. Spectra were obtained by loading the prepared SNP sample into a 1cm path length cuvette.

2.4 Antimicrobial activity

The antimicrobial activity of phytosynthesized AgNPs was assayed against both specific bacterial and fungal pathogens using agar well-diffusion method as given by Perez et al. (26).

Table 1: Scientific and common names of medicinal plants used for the synthesis of AgNP

S.No.	Scientific Name	Family	Common Name	Medicinal Name
1.	Punica granatum	Lythraceae	Anar	For treatment of osteoarthritis, sore throats, coughs, urinary infections, digestive disorders, skin disorders, arthritis, and to expel tapeworms
2.	Datura innoxia	Solanaceae	Datura, Thorn Apple	Antiasthmatic, febrifuge, antipertussive
3.	Aegle marmelos	Rutaceae	Bael	for the treatment of various disorders in human being such as, diabetes, liver toxicity, fungal infection, microbial infection, inflammation, pyrexia and to relieve pain

Table 2: Antibacterial activity of silver nanoparticles and Plant extracts. Values (mean \pm SD) are average of test samples of AgNO₃, AgNPs and plant extracts analyzed individually in triplicate (n = 1X 3)

Name of the test microorganisms	Zone of inhibition (mm) average \pm standard deviation (SD)				
	AgNO ₃	AgNPs	Punica granatum		
Gram positive (Staphylococcus aureus)	5.00±1.00	10.33±1.52	4.65±1.15		
Gram negative Enterobactor aerogens)	9.33±0.57	16.33±1.52	5.67±1.52		
Name of the test microorganisms	Zone of inhibition (mm) average ± standard deviation (SD)				
	AgNO ₃	AgNPs	Aegle marmelos		
Gram positive (Staphylococcus aureus)	7.33±1.52	10.67±1.15	4.33±0.57		
Gram negative Ænterobactor aerogens)	7.67±1.15	19.00±1.00	4.67±0.57		
Name of the test microorganisms	Zone of inhibition (mm) average ± standard deviation (SD)				
	AgNO ₃	AgNPs	Datura innoxia		
Gram positive (Staphylococcus aureus)	9.33±0.57	12.67±1.15	6.33±0.57		
Gram negative (Enterobactor aerogens)	10.33±1.52	15.33±0.57	7.00±1.73		

Table 3: Antifungal activity of silver nanoparticles and Plant extracts. Values (mean \pm SD) are average of test samples of AgNO₃, AgNPs and plant extract analyzed individually in triplicate (n = 1X 3)

Name of the test	Zone of inhibition (mm) average \pm standard deviation (SD)					
microorganisms	AgNO ₃	AgNPs	Punica granatum			
Curvularia lunata	9.33±0.57	10.67±1.00	6.67±0.57			
Aspergillus fumigatus	7.67±1.15	15.33±0.57	5.00±1.00			
Name of the test microorganisms	Zone of inhibition (mm) average ± standard deviation (SD)					
	AgNO ₃	AgNPs	Aegle marmelos			
Curvularia lunata	6.33±0.57	15.67±0.57	4.33±0.57			
Aspergillus fumigatus	9.33±0.57	19.00±1.00	6.00±1.00			
Name of the test microorganisms	Zone of inhibition (mm) average ± standard deviation (SD)					
	AgNO ₃	AgNPs	Datura innoxia			
Curvularia lunata	8.67±1.52	16.67±1.15	4.67±1.15			
Aspergillus fumigatus	7.33±1.52	19.67±0.57	5.67±1.52			

2.4.1 Antibacterial

Gram positive (*Staphylococcus aureus*) and Gram negative (*Enterobactor aerogens*) bacteria were used in the antimicrobial assay. The fresh inoculums of Gram positive and Gram negative bacteria was prepared by growing a single colony overnight in nutrient broth at 35°C. The optical densities (OD) of incubated bacteria in nutrient broth were recorded in UV spectrophotometer (Shimadzu, Japan) at wavelength 600 nm. The optimal desired standard optical density (OD) was obtained as 0.45–0.55, attaining the turbidity to 0.5 McFarland standards (108 CFU/mL). The turbidity was set to the desired range by dilution or otherwise again incubation done if OD were less than the standards values.

The sterilized solid Nutrient agar medium (NAM) plates were swabbed uniformly with bacterial pathogens. In the experiment, all chemicals used were of commercial grade and purchased from Merck, Germany). Circular well of 6 mm diameter were punched in NAM agar plates using a sterile cork-borer. The wells were filled with 10-50µl of AgNO₃, AgNPs and Plant extracts. The plates were incubated at 37°C for 24h. After incubation period, the diameters of zone of inhibition produced with different organisms were measured in (mm) and recorded by using Zonal scale.

2.4.2 Antifungal

The antifungal activity of AgNPs was assayed Curvularia lunata and Aspergillus against fumigatus by using agar well-diffusion technique on potato dextrose agar (PDA) as given by Talibi et al (27). The 100µl of fungal species suspension was inoculated onto the solid of AgNO₃, 10-50 µl of AgNPs and plant extracts. The plates were incubated at (28±2) °C for 72h. Evaluation for each test extract form was done in 3 replicates. After incubation for required period, the zone of inhibition produced were recorded. A clear zone of inhibition observed around the well gives indications of antifungal activity for respective used extract and were measured in (mm) by using Zonal scale.

2.5 Statistical analysis

Statistical analysis was carried out using SPSS Version 16 for the assessment of mean comparisons. The results expressed in terms of mean \pm standard deviation. All data presented are obtained from mean values of triplicate measurements (n = 3), obtained from three individual runs.

3 Results and Discussion

SNPs with their unique properties are proving to be an excellent method for the generation of new effective antibacterial agents. The synthesis of nanoparticles using plants material is an alternative way in which the important secreted biomolecules serves both as reducing and capping agent during the reaction. It is considered as green chemical process as it minimizes the use of toxic chemical agents (28.29). The plant-mediated synthesis of metal nanoparticles has shown to produce nanoparticle of shape and size, which is comparable with those which are produced through physical and chemical techniques (30). In order to achieve this target, researchers have paid their interest towards green synthesis of nanoparticles because of its eco-friendly route (31).

A study on the green synthesis of AgNPs by using Punica granatum, Aegle marmelos and Datura innoxia were carried. As the leaf extracts were mixed with the aqueous solution of silver nitrate, color changes from yellowish orange to brown or blackish, nanoparticles (32), which indicated the synthesis of silver nanoparticles. This color change was due to the reduction of Ag⁺ into Ag, which indicates the formation of silver nanoparticles. The reduction rate and formation of nanoparticles can be amplified further by increase in temperature (33). Similar observations of the colour change in the colloidal solution were reported, indicating the formation of AgNPs, due to the result of excitation of surface plasmon phenomena in SNPs (34).

3.2 UV-Vis Spectra Analysis

UV-Vis spectroscopy is the preliminary technique for the characterization of the silver nanoparticles. It is an indirect method generally used to examine the bio-reduction of SNPs from aqueous AgNO₃ solution. The UV-vis spectroscopy is generally a technique usually employed for characterizing the synthesized SNPs owing to the surface Plasmon phenomenon Morphology (SPR). nanoparticles is very much influenced by the SPR, as it is the basis for measuring adsorption of material onto the surface of metal nanoparticles (35). The reduction of AgNO₃ to SNPs on addition of different plant extracts was confirmed by observing colour change and UV-vis spectrum. The AgNO₃ solution was prepared, added to the plant extract samples and allowed for heating at 90°C for 10 min. The color change was observed due to the reduction of Ag⁺ to Ag⁰ due to the presence of various types of biomolecules in the plant extract (36). The absorption spectra of AgNPs of three various plant extracts were recorded using UV-Vis spectroscopy. The UV-Vis absorption spectra displayed the characteristic absorption peak at a wavelength of 203 nm, 204.5 nm and 197.5 nm for Aegle marmelos, Punica granatum and Datura innoxia, respectively due to surface plasmon resonance (SPR) typical characteristic of AgNPs having kmax values which was reported earlier also in the visible range of wavelength 450–460 nm by Kreibig et al., (36). The sample was analyzed

after centrifuging and redispensing in deionized water. The SPR band indicates the presence of spherical silver nanoparticles in the solution. It is reported that the SPR bands are influenced from shape, morphology, various size, composition as well as also dielectric constant of the synthesized nanoparticles. These results corroborate the findings of Rastogi and Arunachalam (37) for SNPs synthesis using aqueous garlic extract (kmax =414) under sunlight irradiation. Suman et al., (38) reported similar results for the SNPs synthesis using the root extract of *Morinda citrifolia* (kmax = 413 nm). Previous studies have also shown that AgNPs at around 400 nm gives rise to the absorption bands in UV-Vis spectrum (39).

3.3 Antimicrobial activity

The silver nanoparticle solution synthesized by green method has shown highly toxic effect against Gram positive and Gram negative bacteria as well as fungal pathogens. The use of colloidal silver as antibacterial agent has been reported from ancient Greece (40). Silver nanoparticles are found to be very effective micro-organisms because of extremely high surface area. Several scientists are of opinion that silver ions, released from the surface are responsible for AgNPs antibacterial activity (41). In present study, the antimicrobial activity of synthesized plant extracts SNPs was tested against gram positive (Staphylococcus aureus) and gram negative (Enterobactor aerogens) bacterial and fungal *Curvularia lunata* and Aspergillus fumigatus using agar well-diffusion method. All tests were repeated thrice for each treatment and all plant-mediated SNPs did show inhibition zone against all studied bacterial and fungal pathogens.

3.3.1 Antibacterial activity

In the present study, the efficacy of Ag nanoparticles (AgNPs) produced from Punica granatum, Aegle marmelos and Datura innoxia aqueous leaf extract was studied against both gram positive (Staphylococcus aureus) and gram negative (Enterobactor aerogens) bacteria. The average antibacterial activity of synthesized AgNPs against bacterial strains ranged from 19.00 mm to 10.67 mm (the zone of inhibition) (Table 2). The silver nanoparticle synthesized using Datura innoxia showed maximum antibacterial activity for Staphylococcus aureus with 12.67 mm zone of inhibition and minimum activity 10.33 granatum mm was observed in Punica synthesized AgNPs. The maximum activity against Enterobactor aerogens was resulted in Aegle marmelos synthesized AgNPs with 19.00 mm zone of inhibition while Datura innoxia AgNPs showed minimum inhibitory activity15.33 mm.

diameter of zone of inhibition measured from Punica granatum synthesized AgNPs was 10.33 mm and AgNO₃ solution and plant extract showed 5.00 mm and 4.65 mm respectively for Staphylococcus aureus. Indeed, Enterobactor aerogens showed 16.33 mm, 9.33 mm and 5.67 mm zone of inhibition respectively for AgNPs, AgNO₃ solution and Plant extract. The AgNPs synthesized using Aegle marmelos exhibited 10.67 mm zone of inhibition while 7.33 mm and 4.33 mm were observed respectively for AgNO₃ solution and Plant extract against *Staphylococcus aureus*. In fact, in the case of Enterobactor aerogens, it was observed 19.00 mm, 7.67 mm and 4.67 mm respectively for phytosynthesized AgNPs, AgNO3 solution and plant extract. AgNPs synthesized from Datura innoxia showed 12.67 mm while 9.33 mm and 6.33 mm diameter of zone of inhibition was observed for AgNO₃ solution and plant extract respectively against Staphylococcus aureus. And for Enterobactor aerogens 15.33 mm, 10.33 mm and 7.00 mm zone of inhibition was observed for AgNPs, AgNO₃ solution and Plant extract respectively.

In the present study, it was observed that synthesized AgNPs exhibit higher antimicrobial activity in comparison to AgNO₃ and plant extracts. Reddy et al., (41) in his work suggested the less antibacterial activity of aqueous P. longum fruit extract (PLFE) in comparison to the green synthesized silver nanoparticles (PLAgNPs). The enhanced antibacterial activity of SNPs is due to the silver cations released from AgNPs that act as reservoir for the Ag⁺ bactericidal agent. The weak DNA replication process and proteins inactivation are the important mechanisms for antibacterial properties of synthesized metallic nanoparticles (42). Ag⁺ strongly interacts with thiol group of fundamental enzymes and inactivates the enzyme activity (43). Experimental evidence indicates that DNA replication ability affects once the bacteria have been treated with silver ions (43-44). The accurate mechanism of the antibacterial effect of Ag⁺ is partially understood, however, literature survey reveals that the positive charge on the Ag ion is vital for its antimicrobial activity (44,45).

effect is probably antibacterial resulting through the electrostatic attraction between negative charged cell membrane of microorganism and positive nanoparticles. Pal et al. (43) studied antibacterial activity against E. coli, S. aureus and Salmonella typhi and observed that the effect was dose dependent and was more evident against gramnegative compared to gram-positive bacteria. Kumar et al., (46) reported an in situ green biogenic synthesis of gold nanoparticles (AuNPs) using Terminalia chebula aqueous extracts as reducing and stabilizing agent and showed towards positive efficient activity gram Staphylococcus aureus compared to gram

negative *E. coli* using standard well diffusion method.

3.4 Antifungal activity

In the present work, inhibition activity were fungal against evaluated pathogens statistically analyzed. Phytosynthesized AgNPs showed higher antifungal activity in terms of inhibition zone followed by AgNO3 and with only plant extracts represented in Table 3. The antifungal activity of silver nanoparticles (AgNPs) produced from Punica granatum, Aegle marmelos and Datura innoxia aqueous leaf extract was studied against Curvularia lunata and Aspergillus fumigatus. The average antifungal activity of synthesized AgNPs against fungal pathogens ranged from 19.67 mm to 10.67 mm (the zone of inhibition) (Table 3).

Among the tested SNPs against selected fungal pathogens, AgNPs synthesized from Datura innoxia found to have higher antifungal activity against Aspergillus fumigatus with 19.67 mm where as Punica granatum AgNPs showed minimum activity with 15.33 mm (Zone of inhibition). In case of Curvularia lunata, it was observed that Datura innoxia synthesized AgNPs showed maximum 16.67 mm and Punica granatum exhibits minimum 10.67 mm zone of inhibition. The diameter zone of inhibition measured for phytosynthesized Punica granatum, Aegle marmelos and Datura innoxia AgNPs were 10.67 mm, 15.67 mm and 16.67 mm respectively against *Curvularia lunata*. AgNPs synthesized using *Punica granatum*, *Aegle marmelos* and Datura innoxia showed 15.33 mm, 19.00 mm and 19.67 mm zone of inhibition against A. fumigatus. The zone of inhibition observed for AgNO₃ and *Punica granatum* plant extract was 9.33 mm and 6.67 mm against Curvularia lunata while 7.67 mm and 5.00 mm showed for Aspergillus fumigatus. Curvularia lunata showed 6.33 mm and 4.33 mm zone of inhibition for AgNO₃ and Aegle marmelos extract respectively. The zone of inhibition was 9.33 mm and 6.00mm for AgNO₃ and Aspergillus marmelos extract against fumigatus. Diameter zone of inhibition exhibited for AgNO₃ was 8.67 mm and for *Datura innoxia* extract 4.67 mm against Curvularia lunata. For Aspergillus fumigatus, the zone of inhibition was 7.33 mm and 5.67 mm respectively in the case of AgNO₃ and Datura innoxia extract.

AgNPs synthesized using *Datura innoxia* showed maximum antifungal activity with 19.67 mm while *Punica granatum* synthesized SNP exhibited 15.33 mm against fungal pathogens *Aspergillus fumigatus*. For *Curvularia lunata, Datura innoxia* aqueous extract synthesized AgNP give maximum inhibitory activity of 16.67 mm and minimum 10.67 mm zone of inhibition was observed for *Punica granatum* extract synthesized AgNPs. Researchers have reported antifungal activities of silver nanoparticles (47). There are reports of several fungal strains showing effective drug resistance like

Fusarium solani, Candida albicans, Aspergillus flavus and Candida glaberata (48).

4 Conclusion

Plants or their extracts can be resourcefully used in the synthesis of silver nanoparticles with vast applications. Control over the shape and size of nanoparticles seems to be very simple with the use of plants. The nanoparticle synthesis using plant extract provides with acceptable morphology as well as size of nanoparticles. This approach for the synthesis of silver nanoparticles has many advantages such as, ease with which the process can be scaled up in affordable way, economic viability, and eco-friendly. The present study supports the medicinal values of these plants and also demonstrated a simple, rapid and economical route for the synthesis of silver nanoparticles; and their potential of rendering antimicrobial efficacy. Moreover the synthesized SNPs are highly stable and also reproducible, thus results in the improvement of the therapeutic effectiveness as well as emphasizes the medicinal values of these plants.

Author Contribution Statement

RD conducted the experiment and carried out the statistical analysis, KS designed and structured the experiment, and BG structured and wrote the manuscript.

Conflict of Interest

The author declares that she has no conflict of interest.

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