

RESEARCH ARTICLE



Investigating the antioxidant, anti-inflammatory and anticancer potential of *Ononis natrix* L. with phytochemical analysis

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Abstract

Ononis natrix L. (O. natrix) belongs to Fabaceae family and grows on the coast of Mediterranean Sea. The current study aims to investigate the antioxidant, anti-inflammatory and anti-cancer activities of the methanolic extract of O. natrix and identify the phytochemical constituents responsible for these effects. Gas chromatography-mass spectrometry (GC-MS) was employed to characterize the phytochemicals found in the extract. The human red blood cells (RBCs) membrane stabilization method, ferric reducing antioxidant power (FRAP), total antioxidant capacity (TAC), 1,1-diphenyl -2-picryl hydrazyl (DPPH) radical scavenging assay and MTT viability assay were used to investigate the anti-inflammatory, antioxidant and anti-cancer activities. The results showed that methyl stearate (39.48%) and hexadecanoic acid, methyl ester (35.61%) were the most relatively abundant phytochemicals identified by GC-MS. The percentage of hemolysis inhibition for RBC membrane achieved by *O. natrix* extract was 99.6 ± 1% versus that by indomethacin was $99.9 \pm 0.6\%$. Additionally, the DPPH radical scavenging activity of O. natrix extract versus ascorbic acid was 97 ± 0.3% and 98 ± 0.2%, respectively. The O. natrix extract showed a moderate anti-cancer activity against DLD-1 cells (IC₅₀ = 57.13 \pm 4.9 µg/mL), MCF-7 cells $(IC_{50} =$ 96.69 \pm 6.2 µg/mL) and A549 cells (IC₅₀ = 126.63 \pm 10.4 µg/mL), when compared with the anti-cancer drug; Cisplatin. Therefore, O. natrix extract showed significant anti-inflammatory, antioxidant and anti-cancer properties, which could be attributed to methyl stearate and hexadecanoic acid, methyl ester constituents.

Keywords

anti-inflammatory; antioxidant; cancer; GC-MS; O. natrix

Introduction

Several anti-inflammatory medications such as ibuprofen, ketoprofen and ketorolac show serious adverse effects, including stomach and duodenum ulcers, bleeding hypertension, nephrotoxicity and ototoxicity (1-4). However, they are currently being used widely due to their over-the-counter availability (5). Additionally, anti-inflammatory medications are being prescribed by different medical specialists for several diseases as painkillers (6). Different plants have been investigated to find phytochemicals with anti-inflammatory potencies and lower adverse effects. For example, coffee leaf

extract was shown to reduce the levels of proinflammatory cytokines like tumor necrosis factor- α (TNF- α), interleukin (IL)-1 β and IL-8; which were induced by lipopolysaccharide treatment of co-cultured colorectal adenocarcinoma (Caco-2) and myeloid leukemia (U937) cell lines (7).

Oxidative stress has been linked with cardiovascular and neurodegenerative diseases, such as hypertension (8), myocardial infarction (9), atherosclerosis (10), Alzheimer's disease (11) and Parkinson's disease (12). Oxidative stress is defined as a disproportion between elevated reactive oxygen species (ROS) and inadequate cellular antioxidant activity. Elevated oxidative stress can harm cellular structure and even cause tissue destruction (13). Remarkably, the antioxidant enzymes such as catalase, glutathione peroxidase and superoxide dismutase are responsible for scavenging the oxidative free radicals from the cell (14). Several plant species have been explored for their antioxidant properties with the isolation of antioxidant phytochemicals from different plant extracts. For example, Tephrosia purpurea (L.) subsp. apollinea extract showed a potential antioxidant activity (15).

Cancer is the second primary cause of death worldwide, with 10 million deaths reported annually (16). Cancer is characterized by abnormal and uncontrolled rapid cell growth affecting any part of the human body. It is viewed as a serious disease that requires consistent efforts for drug discovery (17). Interestingly, more than half of the anti-cancer drugs are isolated from plants. These medicinal plants mostly contain several polyphenolic compounds which inhibit cancer cell growth (18).

There are 20000 species in the Fabaceae family. The genus Ononis, a member of the Fabaceae family, has 17 species. The majority of those species are found in Middle Eastern and Mediterranean regions (19). One of the species, O. natrix, is primarily found in Tunisia, Libia, Egypt, Palestine, Lebanon and Jordan (20). Remarkably, various chemical constituents of O. natrix have shown different therapeutic properties. For instance, 3,4-dihydroisocoumarin derivatives and 5-alkylresorcinol extracted from O. natrix acetonitrile sub-fraction of acetone showed moderate anti-microbial and anti-protozoal activities against several protozoa and microorganism strains (21). In another study, 21 components, including 6-(2'R-acetoxypentadecyl)-2-hydroxy-4-methoxybenzoicacid isolated from O. natrix were analyzed for anti-bacterial, anti-leishmanial and anti-trypanosomal properties (22). Quercetin, kaempferol and flavones were among the other chemical components discovered in the O. natrix leaf extract which exhibited broad spectrum anti-microbial activities (23). O. natrix has been renowned for its potential anti-microbial (23), anti-Alzheimer, anti-diabetic (20), antioxidant (24) and anti -hypertensive properties (24). Other studies demonstrated significant biological activities of Ononis species, including wound healing, cytotoxicity, anti-spasmodic and analgesic properties (25). Therefore, several Ononis species have been traditionally used for the treatment of gout, diabetes, renal and skin disorders (25). Ononis species, either alone or in combination with other diuretic plants, have been

marketed as herbal teas (26). Very recently, *O. natrix* was found to have anti-proliferative effects against hepa-tocellular carcinoma and prostate cancer (20).

This study aims to investigate the antioxidant, antiinflammatory and anti-cancer activities of the methanolic extract of *O. natrix* and identify the phytochemical constituents responsible for these effects.

Materials and Methods

Plant materials

During spring, the *O. natrix* aerial parts were assembled from the City of Mutah, Al-Karak governorate, Jordan. The plant was cleaned using tap water and was dehydrated by incubation at 25°C, for 10 days in a dark, ventilated room. A grinder (Moulinex, France) was employed to triturate the dried plant (15). A descriptive list was used to identify the plant (27). The Biology Department, Faculty of Agriculture, Mutah University, authenticated the plant and voucher specimens (711/5/1998) deposited in the department.

Preparation of crude extract

The cold percolation technique extracted 250 g of powder from the air-dried plant. In three separate containers, the powder underwent agitation for 3 days at 25°C with 70% methanol (500 mL). The methanol extract was then filtered using a Buchner funnel. Subsequently, a rotary evaporator was utilized to eliminate methanol under reduced pressure at 40°C. Traces of solvent were evaporated using a desiccator, resulting in the acquisition of 20 g per 100 g of dry weight of crude extract of *O. natrix*, which was subsequently subjected to chemical compound characterization via Gas chromatography/Mass spectrometry (GC/MS) (15).

Quantitative analysis of chemical compounds by GC/MS

The previously prepared crude extract underwent analysis utilizing a GC/MS (Thermo Scientific TRACE 1310) instrument (J & W Scientific). The grade of the flow of helium was 1 mL/min. The oven temperature was gradually elevated from 40°C to 280°C at a grade of 5°C per min. The quantities of injection were conserved at 1 μ L, with a hold time of 7.5 min. The temperature of the ion source was set to 280°C and the ionization voltage of 70 eV was applied. Samples were ionized in the mode of electron impact using a range of mass of m/z 50–650. Data interpretation was based on database sources from the NIST and Wiley libraries.

Evaluation of anti-inflammatory activity

Human red blood cells (RBCs) membrane stabilization technique

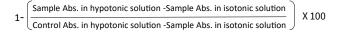
Owing to the similarities between membranes of RBCs and lysosomes, phenylhydrazine (PhNHNH₂), heat or hypotonic medium may cause membrane lysis and content release. Indomethacin prevents lysosomal membrane breakdown and enzyme release. The anti-inflammatory properties of *O. natrix* were assessed by comparing its ability to prevent RBC membrane lysis caused by hypotonic solution and indomethacin as the positive control (28).

Erythrocyte suspension preparation

Fresh whole blood (3 mL) was collected into heparinized tubes from participants who had not taken analgesics for two weeks before the study. A complete blood count (CBC) was conducted to rule out any hematological disorders. The tubes were then centrifuged for 10 min at 3000 rpm. The RBC pellets were solubilized in normal saline and centrifuged again at 3000 rpm for 10 min to wash off the supernatant. This step was repeated thrice and the final RBC pellet was reconstituted using isotonic solution (phosphate buffered saline containing 1.15 g of Na₂HPO₄, 0.2 g of 10 mM NaH₂PO₄, 9 g of NaCl and pH 7.4 in 1 L of distilled water) to give 40% v/v suspension. This cellular suspension was used for the anti-inflammatory assay (29).

Hemolysis induced by hypotonicity

5 mL of the hypotonic solution was used to dissolve 1 g of O. natrix methanolic extract. The centrifuge tubes were loaded with dilutions ranging from 100–1000 µg/mL. One g of the extract was solubilized in 5 mL of the isotonic solution and the extract concentrations (100–1000 µg/mL) were also added to the centrifuge tubes. The control tubes received 5 mL of hypotonic solution and 5 mL of indomethacin (200 µg/mL). The previously prepared suspension of erythrocyte (100 µL) was added to each tube and mixed. The mixtures were incubated for 1 hr at 37°C and then centrifuged at 1300 rpm, for 3 min. The supernatant (hemoglobin) was then collected and its absorbance at 540 nm was recorded using Biosystm 310 plus spectrophotometer. Equation 1 was used to calculate the percentage of hemolysis inhibition induced by O. natrix (29). %Hemolysis inhibition =



Antioxidant activity evaluation

Total antioxidant capacity assay (TAC)

3 mL of reagent solution (28 mM Sodium phosphate (NaH₂PO₄), 0.6 M H₂SO₄ and 4 mM Ammonium molybdate) were mixed with 1 mL of 0.5 mg/mL of each sample of *O. natrix* extract. The mixture was incubated for 2.5 hr at 95°C and post incubation, the samples were cooled to 25°C, followed by recording of absorbance at 630 nm using a microtiter plate reader (Biotek ELX800; Biotek, Winooski, VT, USA). Ascorbic acid (1 mg/mL) was used as the reference drug. Values were represented as ascorbic acid equivalent (AAE) µg/mg of extract (30).

Ferric reducing antioxidant power (FRAP) assay

40 mL of extract samples were mixed with 50 mL 10% trichloroacetic acid, 50 mL (0.2 mol/L) Na₂HPO₄. 2H₂O buffer and 50 mL 1% potassium ferricyanide (K₃Fe (CN)₆) in Eppendorf tubes. At 3000 rpm, the mixture was centrifuged for 10 min. The 96-well plates were loaded with the supernatant (166.66 µL) of each sample and three wells were kept for the negative control, DMSO, followed by the addition of 33.3 µL of ferric chloride (FeCl₃ 1%) to each well. The antioxidant compounds in the extract can donate electrons or hydrogen; therefore, the reduction of Fe³⁺ to Fe²⁺ by these compounds was detected colormetrically using a microtiter plate reader (Biotek ELX800; Biotek, Winooski, VT, USA) at 630 nm. Ascorbic acid (1 mg/mL) was used as the reference drug. The values were expressed as ascorbic acid equivalent (AAE) μ g/mg of extract (31).

1,1-diphenyl-2-picryl hydrazyl (DPPH) assay

The antioxidant potential of *O. natrix* was determined using the DPPH assay. A solution of 0.1 mM DPPH in ethanol was prepared; 1 mL of this solution was combined with 3 mL of *O. natrix* extract at various concentrations (ranging from 3.9–1000 µg/mL). The resulting mixture was vigorously shaken and incubated at 25°C for 30 min before measuring the optical densities at 517 nm employing a UV-VIS spectrophotometer (Milton Roy). Ascorbic acid was used as the reference drug. The experiment was carried out in triplicates. The percentage of DPPH radical scavenging action was computed as follows:

DPPH radicals scavenging (%) =

Absorbance of control - Absorbance of extract samples X100

Absorbance of control	
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3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-2H-tetrazolium bromide (MTT) cell viability assay

The cell lines of human dermal fibroblasts (HDF), human ER-positive breast adenocarcinoma (MCF-7), colorectal adenocarcinoma cell line (DLD-1) and human lung adenocarcinoma (A549) were cultured and maintained in RPMI 1640 medium supplemented with 10% fetal bovine serum, penicillin, streptomycin and L-glutamine. The cells were maintained under standard culture conditions of 95% humidity and 5% CO₂ at 37°C.

For the viability assay, the cells were seeded at a density of 1×10^4 cells in each well of a 96-well plate for 24 hr. *O. natrix* extract was diluted with RPMI 1640 medium to produce serial dilutions from 1–1000 µg/mL, 100 µL of each dilution was added to the corresponding well in triplicates and incubated for four days. After 4 days, 5 mg/mL of MTT solution was added to the treated cells to form formazan products only by the living cells and finally, dimethyl sulfoxide was used to dissolve the purple formazan crystals. Absorbance of the formed color was measured spectrophotometrically by a plate reader at 500–600 nm according to an earlier study (32).

The half-maximal inhibitory concentration (IC_{50}) and the selectivity index (SI) of the *O. natrix* extract were calculated according to the following equations:

$$IC_{50}\% = (100 - OD \text{ of test})/OD \text{ of control} \times 100$$

$$.....(Eqn 3)$$

$$SI = IC_{50} \text{ for } HDF \text{ cells}/IC_{50} \text{ for cancer cells}$$

$$.....(Eqn 4)$$

Microscopic examination

The cells were examined after treatment with *O. natrix* extract under a light microscope (Nikon, 118810) at 400X magnification power) to reveal any cellular abnormalities.

.....(Eqn 1)

.....(Eqn 2)

Results

GC/MS analysis

GC-MS was used to detect the chemical composition of O. natrix extract. GC-MS analyses allowed identification of 23 compounds belonging to the chemical classes of fatty acid esters, fatty acids, phthalate esters, sterols, aromatic compounds, aliphatic ketones and flavonoids (Table 1 and Fig. 1). The major compounds of O. natrix extract were fatty acid esters methyl stearate (Relative abundance (R.A)=39.48%) and hexadecanoic acid, methyl ester (R.A=35.61%). Other fatty acid esters such as octadecanoic acid, ethyl ester (ethyl stearate) (R.A=0.67%), hexadecanoic acid, 14-methyl -, methyl ester (R.A=0.61%), octadecanoic acid,4-hydroxy-, methyl ester (R.A=0.56%), hexadecanoic acid,1-(hydroxylmethyl)-1,2-ethane diyl ester (R.A=0.37%), glyceryl 1,3distearate (R.A= 0.31%), pentadecanoic acid, 14-methyl-, methyl ester (R.A=0.27%) and hexadecanoic acid,2,3dihydroxypropyl ester (R.A=0.27%) were identified in O. natrix extract but in small quantities. The identified fatty acids were octadecanoic acid (Stearic acid) (R.A=1.93%), hexadecanoic acid (R.A=0.71%), oleic acid (R.A=0.40%) and n-hexadecanoic acid (R.A=0.15%). Furthermore, the phthalate esters diisooctyl phthalate (1,2-benzene dicarboxylic acid, dioctyl ester) (R.A=0.34%) and glyceryl 1,3distearate (R.A=0.31 %) were identified in the extract. The detected sterols were cholest-5-en-3-yl myristate (R.A=0.14%), cholesta-4,6-dien-3-ol, (3α)- (R.A=0.26%), Table 1. Phytochemical characterization by GC/MS analysis

cholest-3,5-diene (Oxysterol) (R.A=0.93%), stigmasta-5,22dien-3-ol, acetate, (3α)- (R.A=0.23%) and stigmast-5-en-3-ol, oleate (R.A=0.69%). The plant extract also contained the aromatic amine compound: 1H-purin-6-amine, [(2 fluorophenyl) methyl]- (R.A=0.25%), the flavonoid: 6,8-Di-C-áglucosylluteolin (R.A=0.21%) and the aliphatic ketone: 18pentatriacontanone (R.A=0.39%).

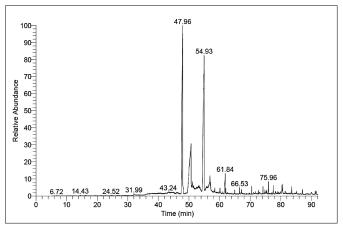


Fig. 1. The methanolic extract GC-MS chromatogram of O. natrix.

Anti-inflammatory potential

The human RBCs membrane stabilization technique was employed to determine the anti-inflammatory effect of *O. natrix* methanolic extract compared to a reference drug, indomethacin. As a result, at the concentrations from

S. No.	Characterized phytochemicals	Weight (g/mol)	Formula	Category	Retention time (min)	Relative abun- dance (%)
1	Hexadecanoic acid, methyl ester	270	$C_{17}H_{34}O_2$	Fatty acid ester	47.96	35.61
2	Pentadecanoic acid, 14-methyl-, methyl Ester	270	$C_{17}H_{34}O_2$	Fatty acid ester	48.17	0.27
3	Hexadecanoic acid	256	$C_{16}H_{32}O_2$	Fatty acid	50.07	0.71
4	n-Hexadecanoic acid	256	$C_{16}H_{32}O_2$	Fatty acid	50.20	0.15
5	Hexadecanoic acid, 14-methyl-, methyl ester	284	$C_{18}H_{36}O_2$	Fatty acid ester	51.10	0.61
6	Hexadecanoic acid, 2,3-dihydroxypropyl ester	330	$C_{19}H_{38}O_4$	Fatty acid ester	51.59	0.27
7	Methyl stearate	298	$C_{19}H_{38}O_2$	Fatty acid ester	54.93	39.48
8	Oleic acid	282	$C_{18}H_{34}O_2$	Fatty acid	55.63	0.40
9	Octadecanoic acid (Stearic acid)	284	$C_{18}H_{36}O_2$	Fatty acid	56.86	1.93
10	Octadecanoic acid, ethyl ester (ethyl stearate)	312	$C_{20}H_{40}O_2$	fatty acid ester	57.04	0.67
11	Hexadecanoic acid, 1-(hydroxymethyl)-1,2-ethane diyl ester	568	$C_{35}H_{68}O_5$	Fatty acid ester	58.46	0.37
12	Octadecanoic acid, 4-hydroxy-, methyl ester	314	$C_{19}H_{38}O_3$	Fatty acid ester	60.13	0.56
13	Glyceryl 1,3-distearate	625	$C_{39}H_{76}O_5$	Fatty acid ester	64.95	0.31
16	Diisooctyl phthalate (1,2-benzene dicarboxylic acid, dioctyl ester)	390	$C_{24}H_{38}O_4$	Phthalate ester	67.21	0.34
17	1,4-Benzenedicarboxylic acid, bis(2-ethylhexyl) ester	390	$C_{24}H_{38}O_4$	Phthalate ester	72.76	0.23
18	Cholest-5-en-3-yl myristate	596	$C_{41}H_{72}O_2$	Sterol	75.20	0.14
19	Cholesta-4,6-dien-3-ol, (3α)-	384	$C_{27}H_{44}O$	Sterol	75.42	0.26
20	Cholest-3,5-diene	368	C ₂₇ H ₄₄	Oxysterol	75.96	0.93
21	Stigmasta-5,22-dien-3-ol, acetate, (3α)-	454	$C_{31}H_{50}O_2$	Sterol	79.90	0.23
22	Stigmast-5-en-3-ol, oleate	678	$C_{47}H_{82}O_2$	Sterol	80.37	0.69
23	1H-purin-6-amine, [(2 fluorophenyl)methyl]-	243	$C_{12}H_{10}FN_5$	Aromatic amine compound	81.43	0.25
24	6,8-Di-C-á-Glucosylluteolin	610	$C_{27}H_{30}O_{16}$	Flavonoids	85.16	0.21
25	18-Pentatriacontanone	506	C35H70O	Aliphatic ketone	91.23	0.39

100–1000 µg/mL, the % hemolysis inhibition of *O. natrix* and indomethacin ranged from 73.8 ± 0.6 to 99.6 ± 1% and 92.5±0.3 to 99.9±0.6%, respectively. Remarkably, there was no statistically significant difference between the % hemolysis inhibition of *O. natrix* and indomethacin at a concentration of 1000 µg/mL (Fig. 2). This suggests that *O. natrix* extract may have an anti-inflammatory effect.

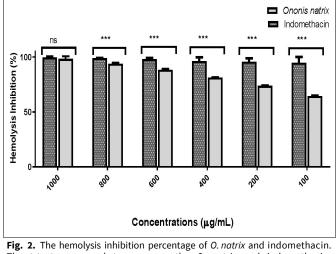


Fig. 2. The hemolysis inhibition percentage of *O. natrix* and indomethacin. The t-test was used to compare the *O. natrix* and indomethacin. "" p = 0.0001 revealed significant differences related to indomethacin, ns: non-significant.

Antioxidant potential

The total antioxidant capacity (TAC) and ferric reducing antioxidant power (FRAP) assays were applied to estimate the antioxidant activity of *O. natrix* compared to a reference drug, ascorbic acid. The ascorbic acid concentrations for TAC and FRAP assays were plotted on the X-axis while the measured absorbances were plotted on the Y-axis (Fig. 3). As a result, TAC and FRAP assays demonstrated 512.59 \pm 0.51 and 411.56 \pm 0.48 µg/mg ascorbic acid equivalent (AAE) of the extract, respectively. This may indicate the potential antioxidant activity of *O. natrix*.

The free radical scavenging was estimated by the DDPH scavenging method. The DDPH scavenging percentage of *O. natrix* was compared to a reference drug, ascorbic acid. The values of IC_{50} of *O. natrix* and ascorbic acid

were computed by GraphPad Prism version 8 (Fig. 4). It was observed that the DDPH radical scavenging percentage of *O. natrix* and ascorbic acid at concentrations from 2–1000 µg/mL were from $38 \pm 0.2\%$ to $97 \pm 0.3\%$ and $47 \pm$ 0.2% to $98 \pm 0.2\%$, respectively. There was no statistically significant difference in the DPPH radical scavenging percentage for *O. natrix* and ascorbic acid at the highest concentration of 1000 µg/mL (p value = 0.5). On the contrary, there was a significant difference between *O. natrix* and ascorbic acid at other concentrations (2–500 µg/mL, pvalue = 0.0001) as shown in Fig. 5. Additionally, the IC₅₀ values for *O. natrix* extract and ascorbic acid were 6 ± 0.2 µg/mL and 3 ± 0.2 µg/mL, respectively. This may suggest that *O. natrix* may possess an antioxidant effect.

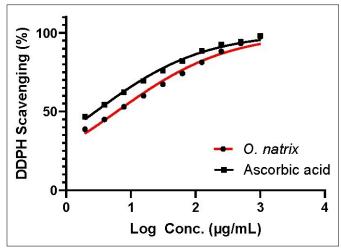


Fig. 4. IC_{50} values estimation of *O. natrix* and ascorbic acid for DDPH scavenging percentage.

Anti-cancer potential

To determine the anti-cancer activity of *O. natrix* extract, the MTT assay was employed to determine the IC_{50} values. The concentrations of the methanolic extract of *O. natrix* were plotted against the % survival of various cell lines (normal and cancerous) such as HDF, MCF-7, DLD-1 and A549. The standard anti-cancer drug, Cisplatin, was used as a positive control and for comparison purposes (Fig. 6 A and B).

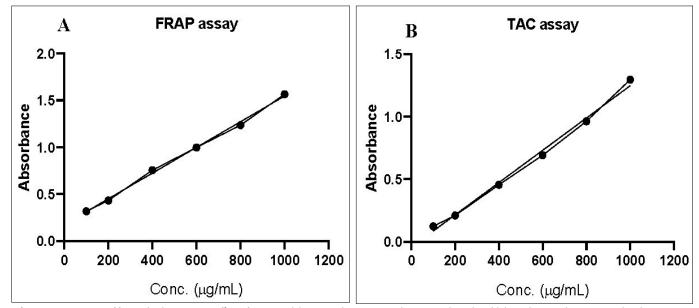


Fig. 3. Determination of free radical scavenging effect of O. natrix. (A) Ferric reducing antioxidant power (FRAP) and (B) Total antioxidant capacity (TAC).

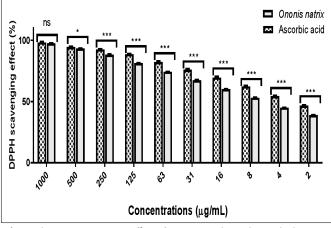
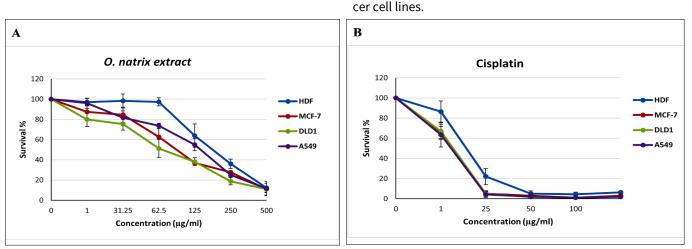


Fig. 5. The DPPH scavenging effect of *O. natrix* and ascorbic acid. The t-test was employed to compare the *O. natrix* and ascorbic acid. \cdots p = 0.0001 illustrates significant differences compared to ascorbic acid, ns: non-significant.



3) (Table 3).

Fig. 6. Determination of the half-maximum inhibitory concentration (IC₅₀) of *O. natrix* extract. (**A**) The concentrations of the methanolic extract of *O. natrix* were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. (**B**) The concentrations of anti-cancer drug; cisplatin were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. (**B**) The concentrations of anti-cancer drug; cisplatin were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. (**B**) The concentrations of anti-cancer drug; cisplatin were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. (**B**) The concentrations of anti-cancer drug; cisplatin were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. (**B**) The concentrations of anti-cancer drug; cisplatin were plotted against the survival % of HDF, MCF-7, DLD-1 and A549 cell lines. HDF: human dermal fibroblasts; MCF-7: human ER positive breast adeno-carcinoma cell line; DLD-1: colorectal adeno-carcinoma cell line.

Table 2 shows the detailed IC_{50} values observed after *O. natrix* extract and cisplatin treatments of cell lines. The results revealed that the anti-cancer activity of *O. natrix* extract was lower than cisplatin with a moderate anticancer activity observed against the DLD-1 colorectal adenocarcinoma cell line ($IC_{50} = 57.13\pm4.9 \ \mu g/mL$) followed by MCF-7 breast cancer cell line ($IC_{50} = 96.69\pm6.2 \ \mu g/mL$) and finally the A549 human lung adenocarcinoma cell line ($IC_{50} = 126.63\pm10.4 \ \mu g/mL$).

Table 2. The $IC_{\scriptscriptstyle 50}$ of O. natrix extract and cisplatin on cancer and non-cancer cell lines

Cell line	IC₅₀ (μg/mL)			
Cettime	O. natrix	Cisplatin		
HDF	202.3 ± 7.3	7.87 ± 2.9		
MCF-7	96.69 ± 6.2	2.35 ± 1.1		
DLD-1	57.13 ± 4.9	1.85 ± 0.7		
A549	126.63 ±10.4	1.78 ± 0.4		

IC₅₀: the half-maximal inhibitory concentration; **HDF**: human dermal fibroblasts; **MCF-7**: human ER-positive breast adenocarcinoma cell line; **DLD**-1: colorectal adenocarcinoma cell line; **A549**: human lung adenocarcinoma cell line.

The methanolic extract of *O. natrix* showed the highest selectivity index (SI) of cytotoxicity against the DLD-1 Table 3. Selectivity index of O. natrix extract against cancer cell lines

	Selectivity index (SI)		
Cell line	0. natrix	Cisplatin	
MCF-7	2.09	3.35	
DLD-1	3.54	4.25	
A549	1.60	4.40	

colorectal adenocarcinoma cell line of 3.54, while negligi-

ble cytotoxic selectivity was observed against MCF-7 or

A549 cell lines (SI = 2.09 and 1.59, respectively). Remarkably, the cisplatin drug showed a prominent cytotoxic selectivity against all three DLD-1, MCF-7 and A549 cell lines (SI>

and A549 cell lines demonstrated normal intact cellular structures and features. However, the *O. natrix* extract-

treated cells exhibited shrunken, necrotic and abnormal

cellular structures. The cisplatin-treated cells showed se-

verely destroyed cells. These results confirm the anti-

cancer activity of O. natrix extracts against a panel of can-

Furthermore, the different cell lines were treated with *O. natrix* extract and examined under a light microscope to reveal the cellular changes following treatment. As shown in Fig. 7, the control cells from HDF, MCF-7, DLD-1

SI: selectivity index; MCF-7: human ER-positive breast adenocarcinoma cell line; DLD-1: colorectal adenocarcinoma cell line; A549: human lung adenocarcinoma cell line. The selectivity index (SI) was calculated by dividing the IC₅₀ value of human dermal fibroblasts (HDF) by the IC₅₀ values for each cancer cell line. The SI values > 3 represent significant anti-cancer agents.

Discussion

In this study, the analysis of GC-MS for *O. natrix* extract showed the existence of several active components, which explains the observed antioxidant and anti-inflammatory properties of *O. natrix*. The fatty acid ester hexadecanoic acid, methyl ester and octadecanoic acid, ethyl ester (ethyl stearate) are known for possessing antioxidant, antiinflammatory and anticancer activities. These compounds are found in plants at the highest concentrations as compared to other identified compounds. Other fatty acid esters such as pentadecanoic acid, 14-methyl-, methyl ester, methyl stearate and hexadecanoic acid,1-(hydroxymethyl) -1,2-ethane diyl ester have been recognized for possessing

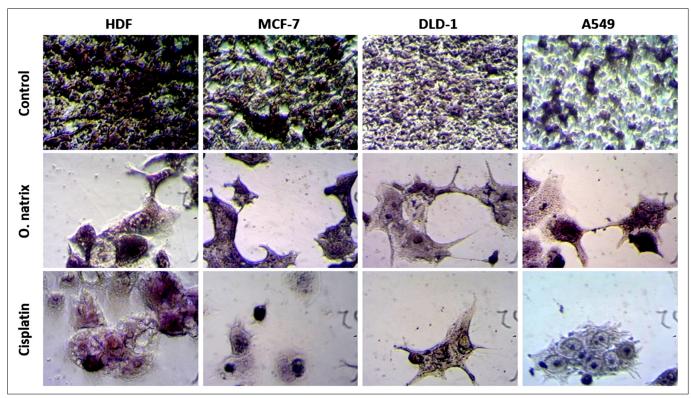


Fig. 7. The anti-cancer effects of *O. natrix* extract against cancer cell lines. Complete monolayer cell sheets from the studied cancer cell lines were examined under the microscope in case of no treatment (control), after *O. natrix* extract treatment and after cisplatin treatment. The control untreated cells from HDF, MCF-7, DLD-1 and A549 cell lines showed normal intact cellular structures and features. However, the *O. natrix*-treated cells exhibited shrunken, necrotic and abnormal cellular structures. The cisplatin-treated cells showed severely destroyed cells.

antioxidant activities. Furthermore, glyceryl 1,3-distearate is renowned for possessing anti-inflammatory activity (33). Additionally, hexadecanoic acid (34), oleic acid (35, 36) and n-hexadecanoic acid (37, 38) are known to have antioxidant, anti-inflammatory and anti-cancer activities. The O. natrix plant contains other compounds that have exhibited antioxidant activities such as 1,4-Benzenedicarboxylic acid, bis (2-ethylhexyl) ester (39) and 1H-purin-6-amine, [(2 fluorophenyl)methyl]- (40). Interestingly, the hexadecanoic acid, 2,3-dihydroxypropyl ester (41), octadecanoic acid (Stearic acid) (42) and di-isooctyl phthalate (1,2-benzene dicarboxylic acid, dioctyl ester) (43) have exhibited antibacterial activities while cholesta-4,6-dien-3-ol, (3α) - has shown antimicrobial activity (44). The plant also contains cholest-3,5-diene which is known to promote wound healing (45).

The O. natrix extract showed a potential antiinflammatory effect in the current study. The findings of the human RBC membrane stabilization method revealed that O. natrix and indomethacin had no significant difference in the inhibition of RBC membrane hemolysis at the concentration of 1000 µg/mL. Additionally, the antiinflammatory effect of O. natrix was enhanced by increasing the extract doses. These observations are compatible with the anti-inflammatory effects of other Ononis species that are closely affiliated with the O. natrix. For example, the *n*-butanol, ethyl acetate and petroleum ether constituents of O. mitissima showed anti-inflammatory and antihemolytic effects at a concentration of 500 μ g/mL (46). However, there are no earlier reports about the antiinflammatory effects of O. natrix, yielding a novel finding for the current study.

The antioxidant assays of FRAP, TAC and DPPH were

used to estimate the antioxidant capacity of O. natrix. The difference in DPPH radical scavenging activities between the O. natrix and ascorbic acid was not statistically significant at the highest concentration of 1000 µg/mL. Therefore, our results show that the potential antioxidant effect of O. natrix is similar to that of ascorbic acid. The current observations were directly compared with the previously reported findings on the antioxidant effect of O. natrix leaf extract which displayed an IC₅₀ value of 29 µg/mL of DPPH radical scavenging ability and a broad spectrum antibacterial effect (23). In the previous study, the methanolic extract of O. natrix leaves was found to have kaempferol, amentoflavone, flavones and quercetin as the main chemical components (23). The extracted essential oils from O. natrix also showed a moderate antioxidant effect (47). The chloroform, ethyl acetate, hexane and methanol extracts from O. natrix leaves showed DPPH radical scavenging ability with IC₅₀ values of 13.45 μ g/mL, 20.03 μ g/mL, 43.92 μg/mL and 49.97 μg/mL, respectively. Therefore, the chloroform extract had a greater antioxidant capacity than ethyl acetate, hexane and methanol (24). Moreover, the ethanol and aqueous extracts from O. natrix demonstrated DPPH radical scavenging ability with IC_{50} values of 80.2 \pm 1.6 µg/mL and 78.1 \pm 0.9 µg/mL, respectively (48). However, in the present study, the IC₅₀ value of DPPH radical scavenging ability is $6 \pm 0.2 \,\mu\text{g/mL}$; indicating antioxidant effect for O. natrix extract stronger than the previously available literature. Therefore, our results confirm the potential antioxidant effect of O. natrix.

The cytotoxic effects of *O. natrix* extract on the PC-3 prostate cancer cell line and HeLa cervical cancer cells were previously assessed using the MTT cell viability assay in which 1 mg of the methanol extract induced apoptosis

YOUSSEF ET AL

in the PC-3 cell line (25). Remarkably, the anti-cancer effect may be related to the occurrence of hexadecanoic acid, methyl ester, octadecanoic acid, ethyl ester and oleic acid found in the O. natrix extract. In a similar study, an ethyl acetate extract of O. natrix showed cytotoxicity against the MDA MB-231 breast cancer cells in a dose-dependent mode, achieving IC₅₀ = $28.75 \pm 2.5 \mu g/mL$. The ethyl acetate extract included terpenes and flavonoids, which may explain the cytotoxic effect (49). Moreover, the cytotoxic effects of O. natrix on other cancer cell lines such as HeLa, MCF-7, CaCo-2, PC-3 and HepG-2 were examined in previous studyand were found variable (20). For example, the O. natrix extract showed a higher cytotoxic effect against HepG-2 and PC-3 cells, when compared to MCF-7, HeLa and CaCo-2 cells. This study is more relevant to our findings, the anti-cancer effect of O. natrix extract against the MCF-7 cells showed an IC₅₀ of 123±2 µg/mL, compared with an IC₅₀ of 96.69 \pm 6.2 µg/mL observed in the current study. Our results confirm the anti-cancer activity of O. natrix as a promising agent for the treatment of various cancer types.

In a recent study, the aerial parts of O. natrix were extracted using methanol, ethyl acetate and water to analyze the phytochemical composition and the various therapeutic activities they posses like antioxidant, anti-Alzheimer, anti-diabetic and tyrosinase activities. The spectrophotometric analysis revealed that the water extract had the highest concentration of total flavonoids (27.88 mg rutin equivalents/g dry weight) while the methanol extract had the highest concentration of total phenolics (74.60 mg gallic acid equivalents/g dry weight). Quantitative phytochemical tests revealed the presence of significant amounts of guercetin, eriodictiol, hyperoside and taxifolin (50). The methanol extract showed the highest activity in all tests, i.e., Phosphomolybdenum, 2,2'-Azinobis-(3-Ethylbenzo-thiazoline-6-Sulfonic acid (ABTS) radical scavenging assay, Copper-reducing antioxidant capacity (CUPRAC) method, Ferric ion reducing antioxidant power (FRAP) assay and metal chelating assays; except for the DPPH radical scavenging activity. The methanol extract also demonstrated greater inhibition of acetylcholinesterase (AChE), butyrylcholinesterase (BChE) and α -glucosidase, while the ethyl acetate extract was most potent in inhibiting α-amylase and tyrosinase activities (50).

Conclusion

The phytochemicals identified in the methanolic extract of *O. natrix* such as hexadecanoic acid, methyl ester, octadecanoic acid and ethyl ester may be responsible for the antiinflammatory, antioxidant and anti-cancer activities. The anti-inflammatory and antioxidant effects of the *O. natrix* were similar to the authentic drugs indomethacin and ascorbic acid. However, the anti-cancer activity of *O. natrix* extract was moderate against DLD-1, MCF-7 and A549 cell lines compared to cisplatin. Therefore, the *O. natrix* extract possesses therapeutic potential against oxidative stress and inflammation, as well as colon, breast and lung cancers with selective cytotoxicity against colon cancer.

8

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Authors' contributions

AY and NE were involved in conceiving and planning the research. AY and MA performed phytochemical analysis by GC/MS and the data acquisition/collection. KA and RH performed the antioxidant assays and calculated the experimental data. MA performed the MTT assay and NE performed anti-inflammatory assay. AY, NE, MA, KA and RH drafted the manuscript and designed the figures. AY, NE, MA, KA and RH aided in interpreting the results. All authors took part in giving critical revision of the manuscript.

Compliance with ethical standards

Conflict of interest: Authors do not have any conflict of interests to declare.

Ethical issues: None

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